

The Absorption Coefficient

$$\frac{I_{\omega}(s_2)}{I_{\omega}(s_1)} = e^{-k_{\nu}(\omega)ds}$$

$k_{\nu}(\nu)$ is the unique *fingerprint* of molecules!

How do we calculate it?

For a single transition
i.e. spectral line

$$k_{\nu}(\omega) = S f(\omega - \omega_0)$$

S is the intensity of the line

$f(\omega - \omega_0)$ is the line shape function

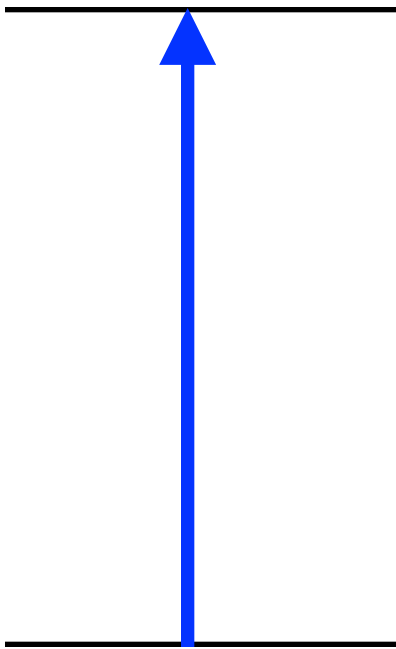
Intensity of a Line

$$\omega = E' - E''$$

$S \propto$ population of lower state
and R^2

R^2 quantum mechanical
probability of the transition
occurring.

The more molecules in
lower state the stronger the
intensity.



Line Shape function

$f(\omega - \omega_0)$ represents the broadening of the line.

γ_D Doppler broadening - due to the motion of the molecules

γ_n natural broadening - due to the natural lifetime of states

γ_L collisional broadening due to collisions of the radiating molecule with other molecules

Line Shape function

Doppler broadening

$$f_D(\omega - \omega_0) = \frac{1}{\alpha_D \sqrt{\pi}} e^{-\left(\frac{\omega - \omega_0}{\alpha_D}\right)^2}$$

where

$$\alpha_D = \frac{\omega_0}{c} \left(\frac{2kT}{m} \right)^{\frac{1}{2}}$$

is the Doppler width of the line

Line Shape function

Lorentz (collisional) broadening

$$f_L(\omega - \omega_0) = \frac{1}{\pi} \frac{\gamma}{(\omega - \omega_0 - \delta)^2 + \gamma^2}$$

Earth's Atmosphere

Dominated by Lorentz broadening.

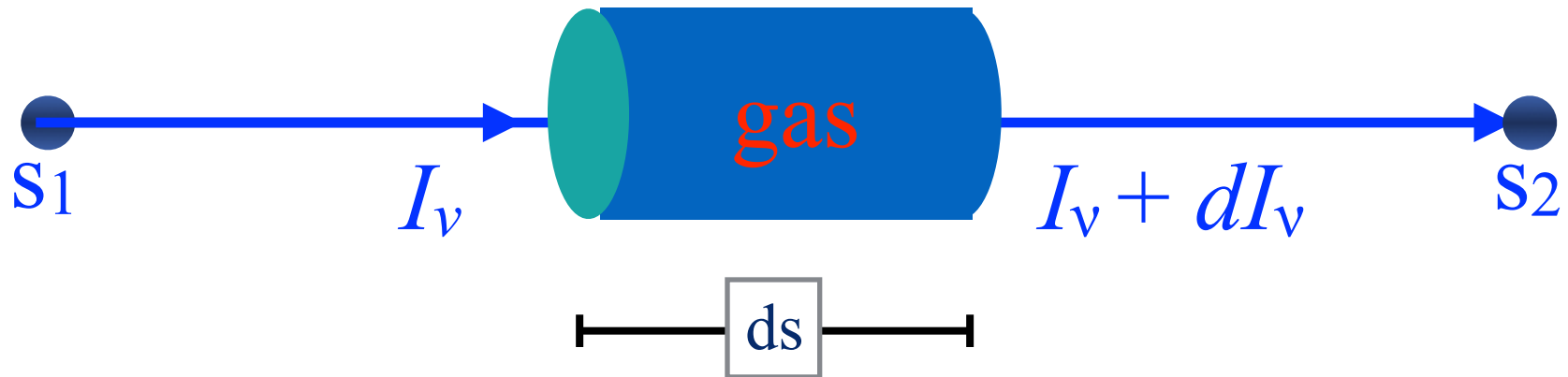
As pressure get low (fewer collisions)

Doppler broadening becomes important.

Voigt function - a convolution of the

Doppler and Lorentz line shape functions.

Single Line



$$I_{\omega}(s_2) = I_{\omega}(s_1) e^{-Sf(\omega - \omega_0) ds}$$

What do we need to do the calculation?

ω, S, γ, E''

Where does one find such information?

HITRAN - www.cfa.harvard.edu/hitran/

~ 5 million transitions

GEISA - ether.ipsl.jussieu.fr/etherTypo/?id=950&L=0

~ 4 million transitions

JPL catalogue - spec.jpl.nasa.gov/ftp/pub/catalog/catdir.html

5 million transitions

High Temperature databases

HITEMP - www.cfa.harvard.edu/hitran/HITEMP.html >122 million transitions

AMES CO₂ - Schwenke *et al.* 2.1 billion transition

CDSD 1000 - Institute of Atmospheric Optics, Tomsk RU, Sergey Tashkun, >3 million transitions

Multiple Lines

$$I_{\omega}(s_2) = I_{\omega}(s_1) e^{-\sum_i S_i f(\omega - \omega_i) ds}$$

sum over lines that influence the absorption at wavenumber ω . Generally $\pm 25 \text{ cm}^{-1}$.

183 GHz line of H2O

ν_1'	ν_2'	ν_3'	ν_1''	ν_2''	ν_3''	J'	K_a'	K_c'	J''	K_a''	K_c''
0	0	0	0	0	0	3	1	3	2	2	0

$\omega = 6.114567 \text{ cm}^{-1}$, $S = 0.779\text{E-}22 \text{ cm}^{-1}/\text{molec}/\text{cm}^2$,
 $\gamma = .0992 \text{ cm}^{-1}/\text{atm}$, $\delta = -.002700 \text{ cm}^{-1}/\text{atm}$ $E'' = 136.1639 \text{ cm}^{-1}$

$$k_v(\omega)ds = S n f(\omega - \omega_0)ds$$

What is the absorption at $6. \text{ cm}^{-1}$ for a 50 cm path at 286 K and 1 atm pressure

$$k_v(\omega) ds = S n \frac{1}{\pi} \frac{\gamma p}{(\omega - \omega_0 - \delta p)^2 + (\gamma p)^2} ds$$

where n is the number density (molec/
 cm^3)

Units

$$S n \equiv (\text{cm}^{-1}/\text{molec}/\text{cm}^2) * (\text{molec}/\text{cm}^3) \\ = 1/\text{cm}^2$$

$$\gamma p \equiv (\text{cm}^{-1}/\text{atm}) * \text{atm} = \text{cm}^{-1}, \omega \text{ is } \text{cm}^{-1}$$

$$\frac{1}{\pi (\omega - \omega_0 - \delta p)^2 + (\gamma p)^2} \quad \text{cm}$$

$$ds = \text{cm}; \text{ so } 1/\text{cm}^2 * \text{cm} * \text{cm} = \text{unitless}$$

number density - Loschmidt's number

$$n_L(p, T) = \frac{n N_a}{V} = \frac{p N_a}{RT}$$

$$n_L(p, T) = \frac{1 \text{ atm} * 6.022 \times 10^{23} \text{ molec} / \text{mole}}{0.082054 \frac{\ell \text{ atm}}{\text{mole K}} * 286 \text{ K} * \frac{1000 \text{ cm}^3}{\ell}}$$

$$n_L(1 \text{ atm}, 286 \text{ K}) = 2.5661 \times 10^{19} \text{ molec} / \text{cm}^3$$

line shape

$$\gamma p = 0.0992 \frac{cm^{-1}}{atm} * 1 atm = 0.0992 cm^{-1}$$

$$\delta p = -.002700 \frac{cm^{-1}}{atm} * 1 atm = -.002700 cm^{-1}$$

$$f(\omega) = \frac{1}{\pi} \frac{\gamma p}{(\omega - \omega_0 - \delta p)^2 + (\gamma p)^2}$$

$$f(\omega) = 1.14250 cm$$

$$k_v(\omega)ds = 0.779 \times 10^{-22} \frac{\text{cm}^{-1}}{\text{molec cm}^{-2}} * 2.5661 \times 10^{19} \frac{\text{molec}}{\text{cm}^3} *$$

$$50 \text{ cm} * 1.14250 \text{ cm}$$

$$= 0.11419$$

$$\frac{I}{I_0} = e^{-k_v(\omega)ds} = 0.892$$

H2O 183 GHz line at Infinite Resolution

