Please email me a script file containing the commands you used to answer these questions at stephen_pennell@uml.edu.

1. Let

$$
A=\left[\begin{array}{lll}
1 & 2 & 3 \\
2 & 3 & 4 \\
3 & 4 & 4
\end{array}\right]
$$

a) Find the determinant of $A$.
b) Find $A^{-1}$.
c) Solve the system

$$
\begin{aligned}
x+2 y+3 z & =2 \\
2 x+3 y+4 z & =2 \\
3 x+4 y+4 z & =1
\end{aligned}
$$

2. Use MATLAB to graph $y=\frac{\sin (2 \pi x)}{1+x^{2}}, y=\frac{1}{1+x^{2}}$, and $y=-\frac{1}{1+x^{2}}$ on the same set of axes for $-1 \leq x \leq 1$.
Please use the following formatting instructions.

- Draw the graph of $y=\frac{\sin (2 \pi x)}{1+x^{2}}$ using a solid blue line, draw the graph of $\frac{1}{1+x^{2}}$ using a dashed red line, and draw the graph of $y=-\frac{1}{1+x^{2}}$ using a dashed green line.
- Create a legend to indicate which curve is which. The only variables in the problem are $x$ and $y$. Don't use other letters in your legend.
- Be sure to label your axes. The only variables in the problem are $x$ and $y$. Don't use other letters in your axis labels.
- Use enough points so your graphs look like smooth curves.

3. A cycloid is specified by the parametric equations $x=r(t-\sin (t)), y=r(1-\cos (t))$.

Draw a cycloid with $r=1.5$ and $0 \leq t \leq 8 \pi$. Use the axis command to make the x axis run from 0 to 40 and the y axis run from 0 to 10 .
4. Generate a figure with a $1 \times 2$ array of windows. In one window draw a loglog plot of the function $C(\omega)=\frac{1}{\sqrt{1+\omega^{2}}}$ for $10^{-2} \leq \omega \leq 10^{3}$, and in the other window draw a plot of $C(\omega)$ with the horizontal axis scaled logarithmically and the vertical axis scaled linearly. Be sure to label the axes. (The string '\omega' will produce the Greek lower case letter $\omega$.)
5. Draw a polar plot of $r=1+\sin (\theta)$ for $0 \leq \theta \leq 2 \pi$.
6. The pressure (in $\mathrm{N} / \mathrm{m}^{2}$ ) of one mole of an ideal gas occupying a volume of $1 \mathrm{~m}^{3}$ is given by $p=8.314 T$, where $T$ is the temperature in degrees Kelvin. The volume (in $\mathrm{m}^{3}$ ) of one mole of an ideal gas at a pressure of $10^{5} \mathrm{~N} / \mathrm{m}^{2}$ is given by $V=10^{5} /(8.314 T)$. Use the plotyy command to graph $p$ and $V$ as functions of $T$ for $300 \leq T \leq 400$. Label the horizontal axis and both vertical axes. Include the units in your axis labels.

