

Self-assembly and Nanotechnology 10.524

Lecture 2. Fundamentals and Theories of Self-Assembly

Instructor: Prof. Zhiyong Gu (Chemical Engineering
& UML CHN/NCOE Nanomanufacturing Center)

Lecture 2: Fundamentals and Theories of Self-Assembly

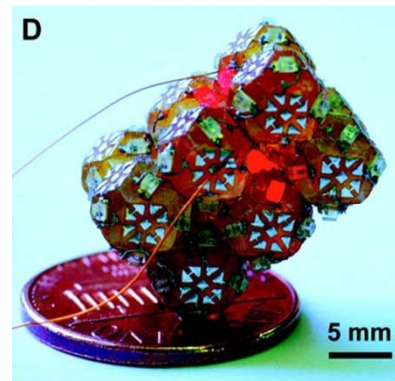
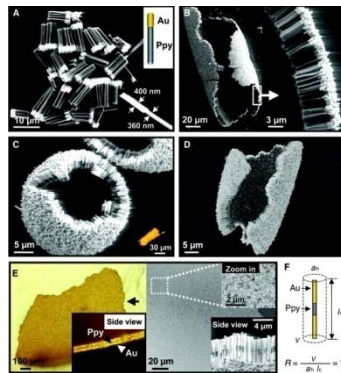
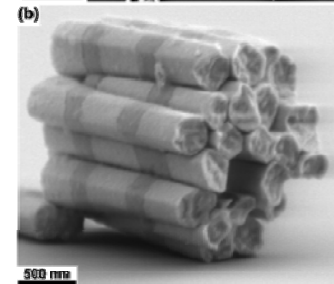
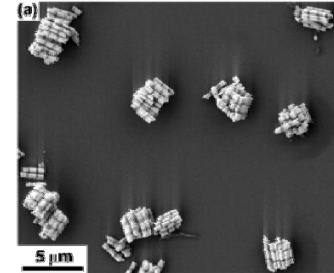
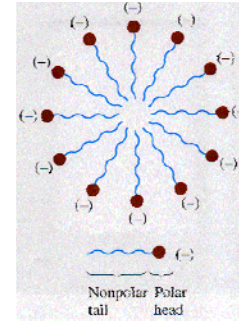
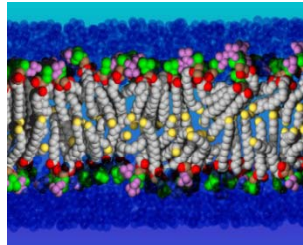
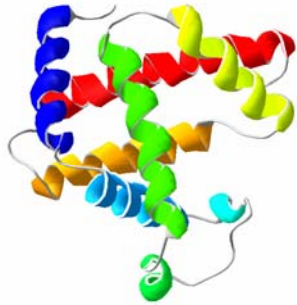
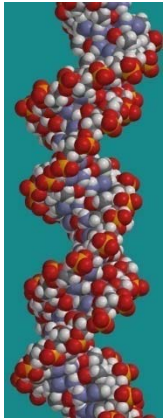
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- Why self-assembly can happen?

Forces, energy, entropy; interactions

- Molecular self-assembly
- Perfect example: DNA self-assembly
- Case study: self-assembly of Block copolymers, measuring the intermolecular forces/interactions
- Surface tension driven (enabled) self-assembly

Fundamentals and Theories of Self-Assembly



- Why self-assembly can happen?
- What are common features?

Forces, energy, entropy

Fundamentals and Theories of Self-Assembly

Various forces and examples

- Covalent bond (chemical bonding)
 - Van der Waals
 - Electrostatic
 - Hydrogen Bonding
 - Hydrophobic
 - Hydration
 - ●●●
- Magnetic force
 - Electrical force
 - Gravity
 - ●●●

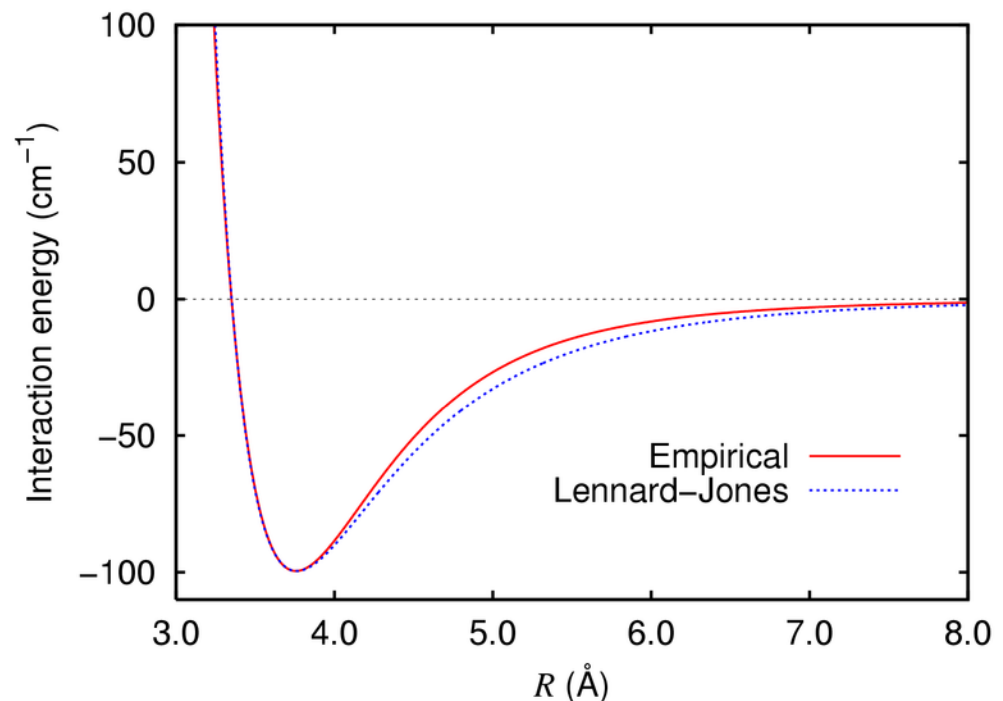
Fundamentals and Theories of Self-Assembly

Van der Waals' forces

Lennard-Jones potential

$$V(r) = 4\epsilon \left[\left(\frac{\sigma}{r} \right)^{12} - \left(\frac{\sigma}{r} \right)^6 \right]$$

where ϵ is the depth of the potential well and σ is the (finite) distance at which the potential is zero.



From Wikipedia

Fundamentals and Theories of Self-Assembly

Hydrophobic force

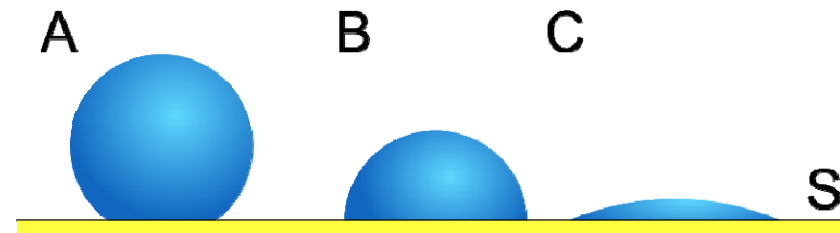
Hydrophobic: Does not like water

Contact angle: $> 90^\circ$



Hydrophilic: like water

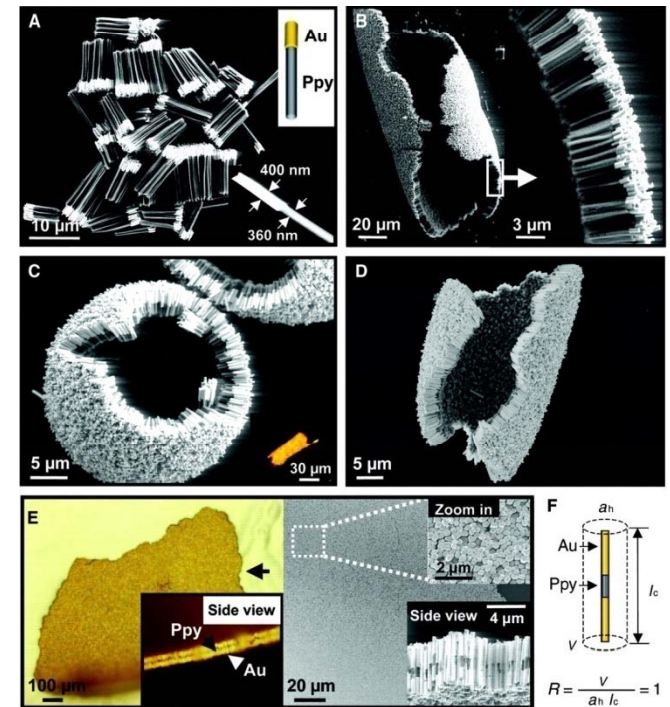
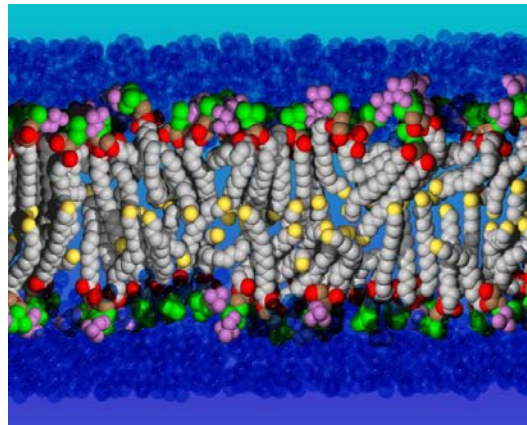
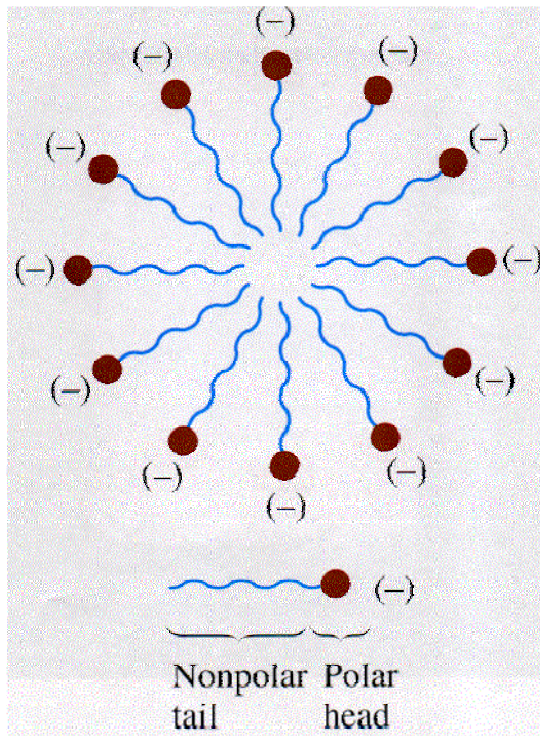
Contact angle: $< 90^\circ$



Wetting of different fluids

Fundamentals and Theories of Self-Assembly

Hydrophobic force: self-assembly examples



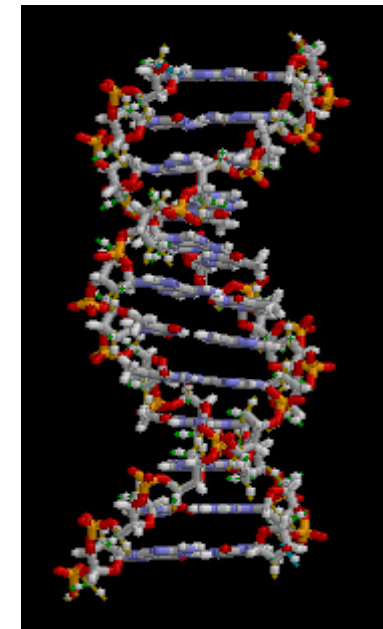
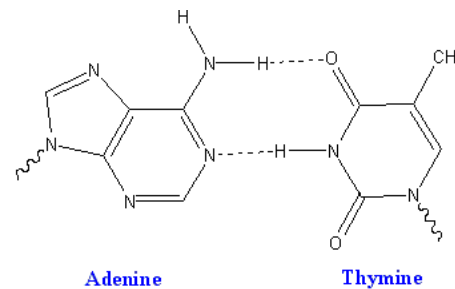
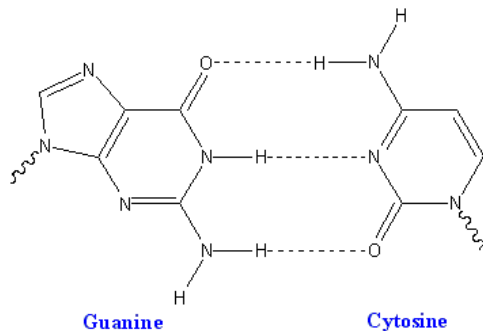
From Wikipedia

Fundamentals and Theories of Self-Assembly

DNA self-assembly and Molecular Computing

Deoxyribonucleic acid (DNA) is a nucleic acid that contains the genetic instructions for the development and function of living organisms.

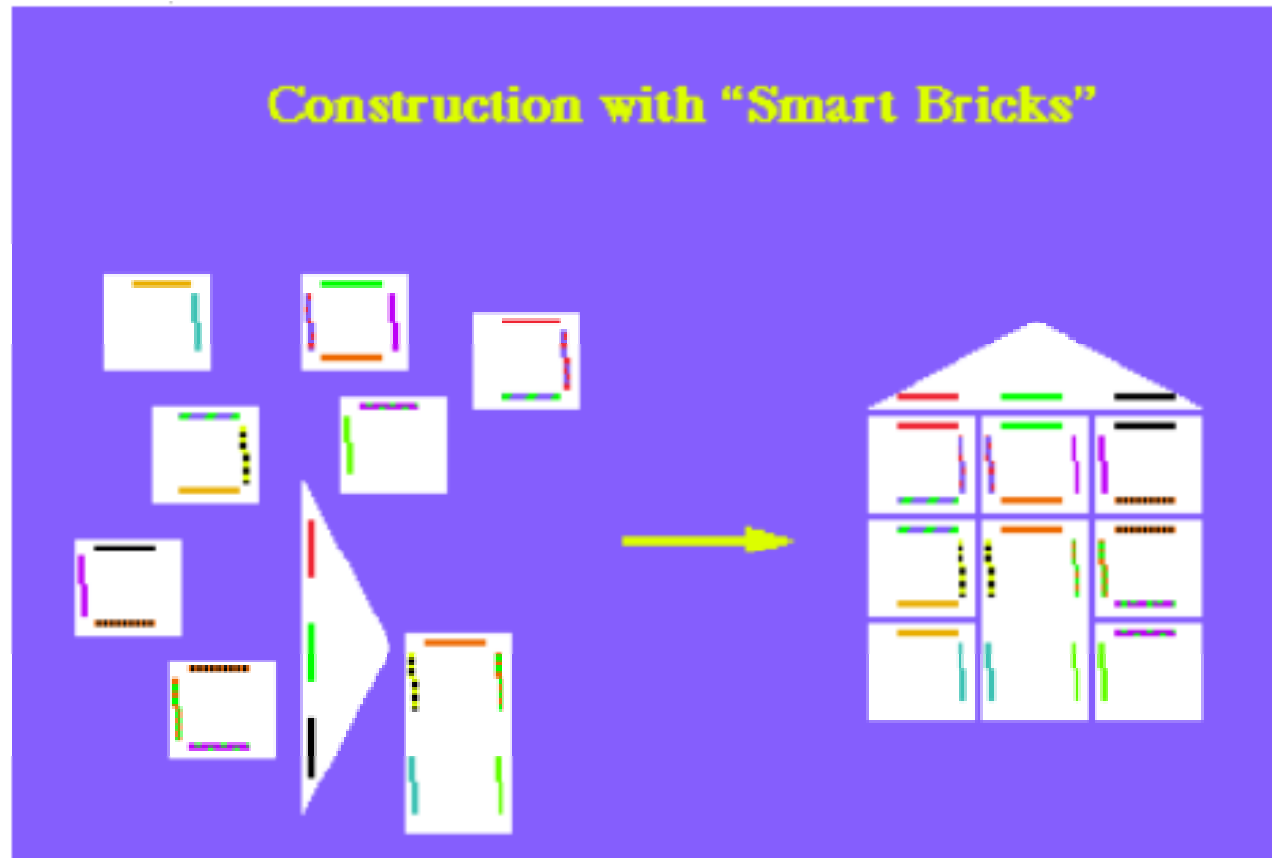
The DNA double helix is held together by hydrogen bonds between the bases attached to the two strands. The four bases found in DNA are adenine (abbreviated A), cytosine (C), guanine (G) and thymine (T).



DNA self-assembly is the most advanced and versatile system known for programmable construction of patterned systems on the molecular scale

Fundamentals and Theories of Self-Assembly

DNA self-assembly and Molecular Computing



Fundamentals and Theories of Self-Assembly

DNA self-assembly and Molecular Computing

An experimental demonstration of an XOR tiling computation based on TAO tiles is reported in [Mao, LaBean, Reif, and Seeman, 00].

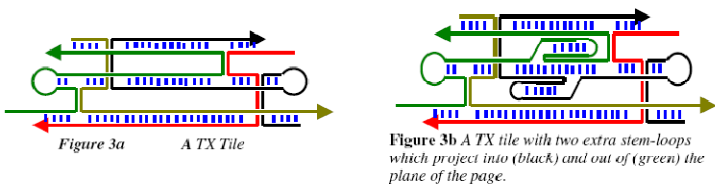
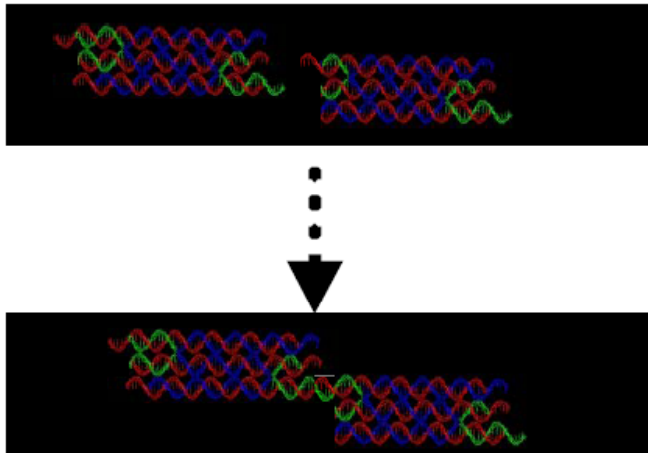
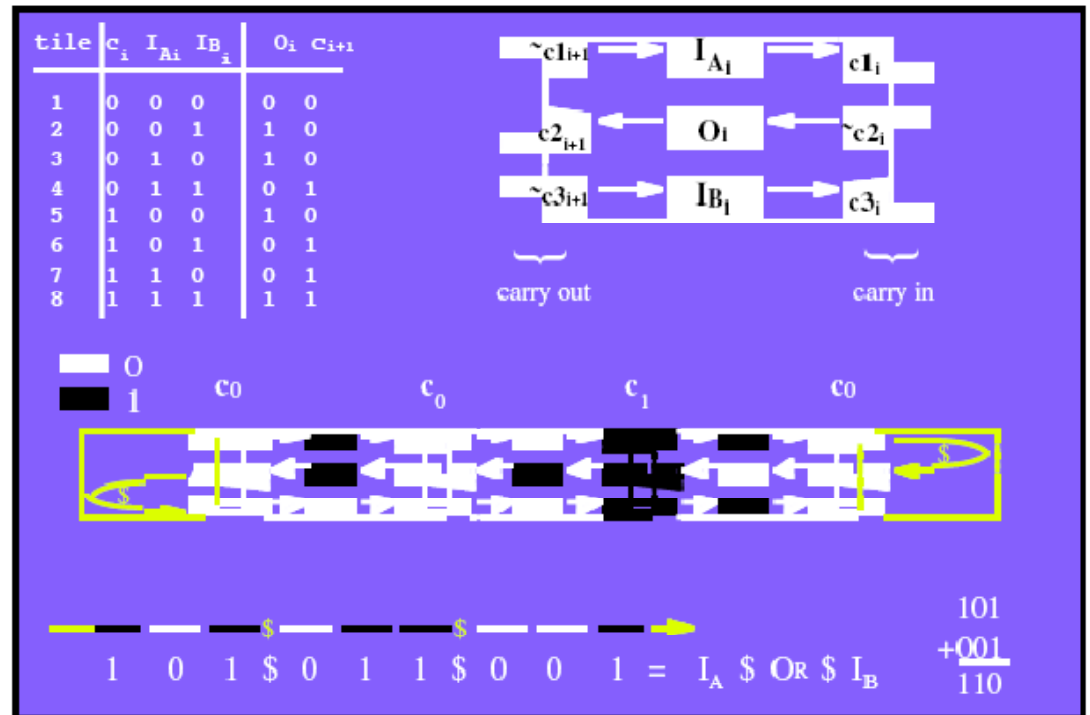


Fig. 3. (a) A triple-crossover tile and (b) a triple-crossover tile that has two extra stem-loops that project into (black) and out of (green) the plane of the page.



The binding of DNA tile pad pairs

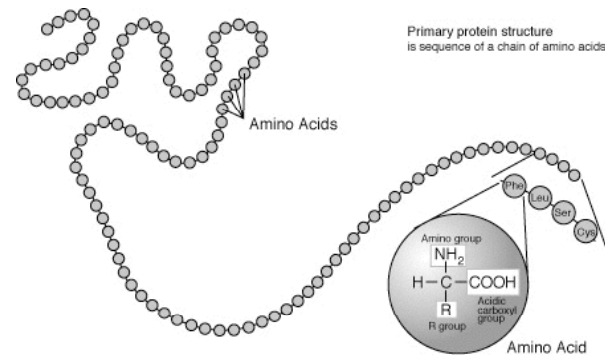


Reif, Duke Univ.

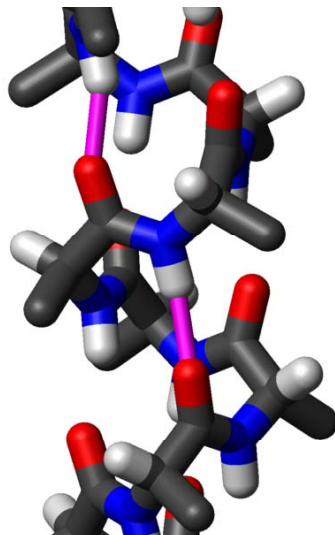
Fundamentals and Theories of Self-Assembly

Protein structures

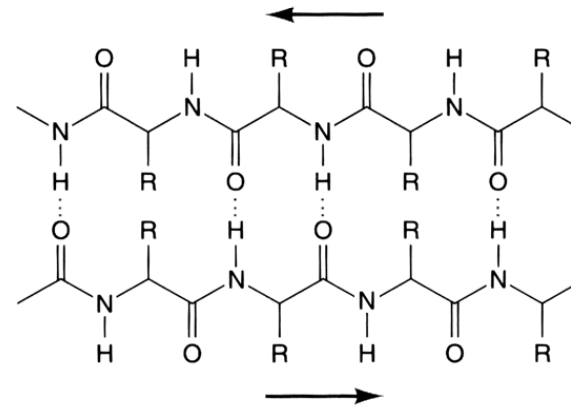
Primary structure: amino acid sequence



Secondary structure: regularly repeating local structures stabilized by hydrogen bonds



alpha helix (α -helix)



β sheet (also β -pleated sheet)

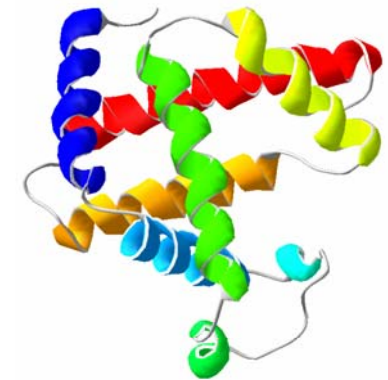
From Wikipedia

Fundamentals and Theories of Self-Assembly

Protein structures

Tertiary structure: the overall shape of a single protein molecule; the spatial relationship of the secondary structures to one another.

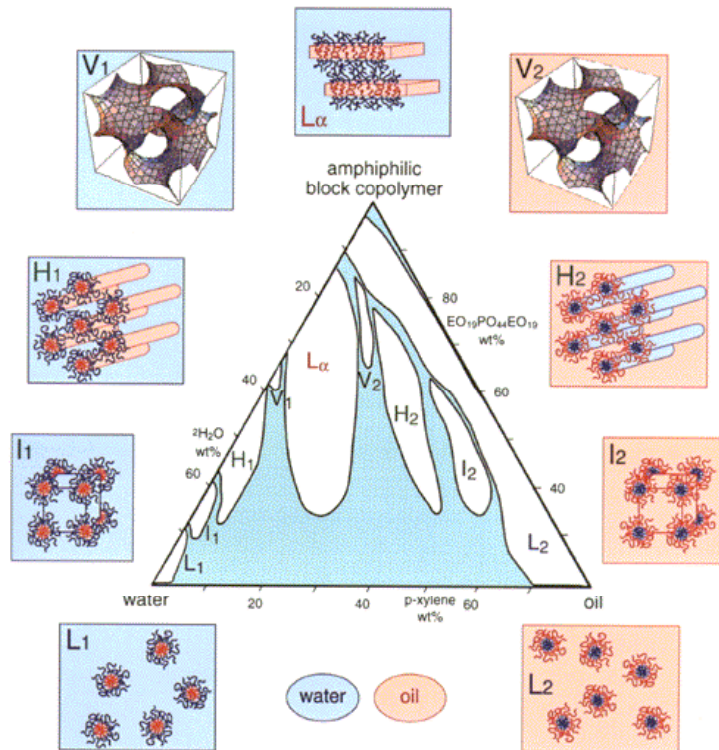
Tertiary structure is generally stabilized by nonlocal interactions, most commonly the formation of a hydrophobic core, but also through salt bridges, hydrogen bonds, disulfide bonds, and even post-translational modifications.



Quaternary structure: the shape or structure that results from the interaction of more than one protein molecule, usually called *protein subunits* in this context, which function as part of the larger assembly or protein complex.

From Wikipedia

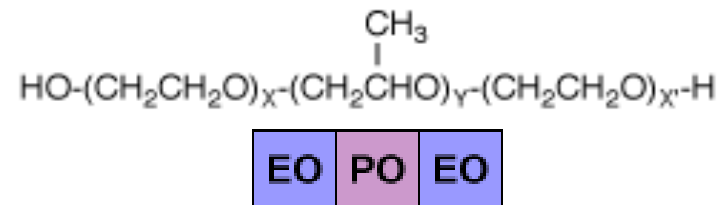
Case Study: Measurement of Intermolecular Forces/Interactions



Ordered nanostructures of block copolymers

Alexandridis, Olsson, Lindman, *Langmuir* **1998**, *14*, 2627-2638

Pluronics®, or Poloxamers, are poly(ethylene oxide) b-poly(propylene oxide)-b-poly(ethylene oxide) block copolymers which consist of two hydrophilic PEO blocks and one hydrophobic PPO block.



PEO-PPO-PEO block copolymers can form a variety of self-assembled nanostructures in water (selective solvent for PEO).

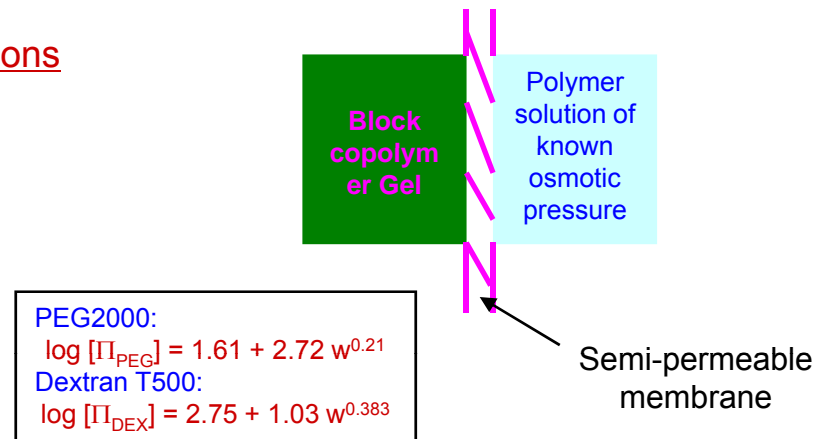
Spherical micelles in solution; micellar cubic, hexagonal, lamellar, bicontinuous lyotropic liquid crystalline phases, etc.

Pluronics® find numerous applications in coatings and personal care formulations, and also in the areas of biomaterials, drug delivery and drug formulations.

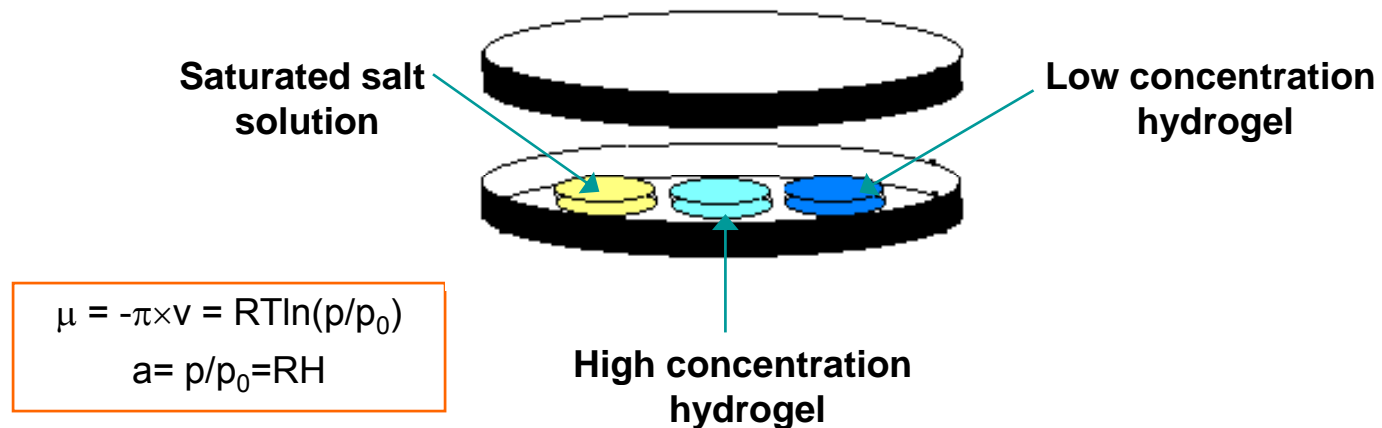
Intermolecular Forces/Interactions in Block Copolymer Assemblies

● Osmotic stress (OS) measurements

I. Equilibration with Polymer solutions

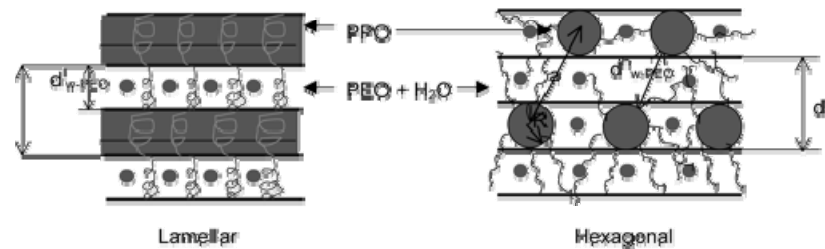
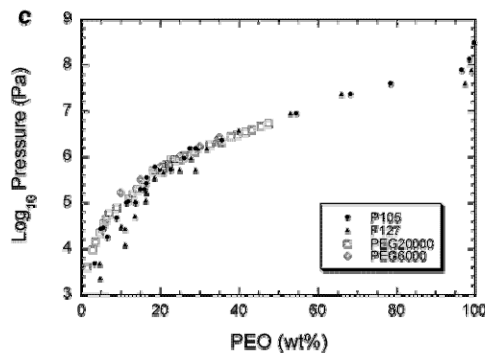
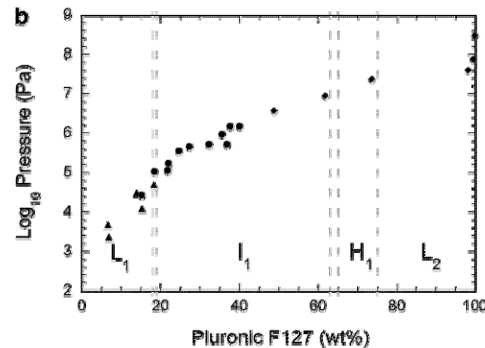
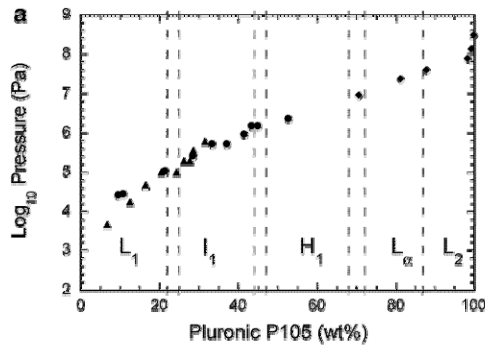


II. Equilibration with saturated salt solutions

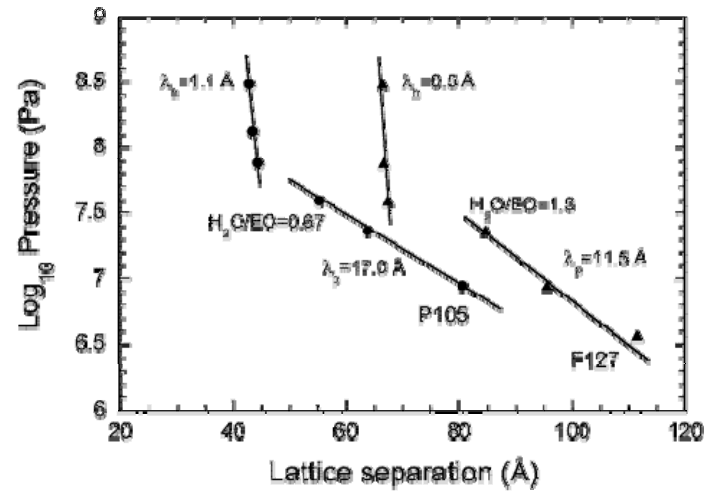


Parsegian, V. A.; et al. *Methods Enzymol.* **1986**, 127, 400-416.

Intermolecular Forces/Interactions in Block Copolymer Assemblies



← SAXS
SANS



- Osmotic pressure is an exponential function of block copolymer concentration. PEO-water interactions contribute more to the osmotic pressure of Pluronic-water systems.
- Different interactions have been observed at different concentration range of Pluronic-water systems.
- Two different decay lengths are observed at force vs. distance curve; one is comparable to PEO coil, another similar to “hydration force”.

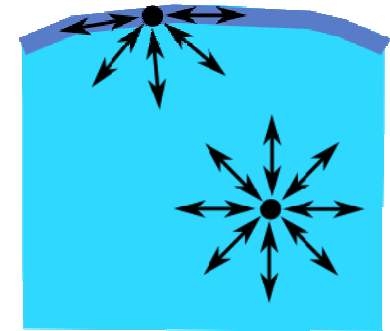
Surface Tension-Powered Self-Assembly

Surface Tension-Powered Self-Assembly

What's surface tension??

In physics, **surface tension** is an effect within the surface layer of a liquid that causes that layer to behave as an elastic sheet

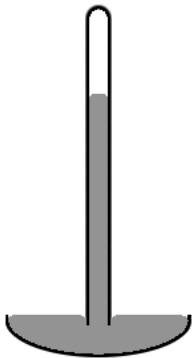
Surface tension is caused by the attraction between the molecules of the liquid by various intermolecular forces



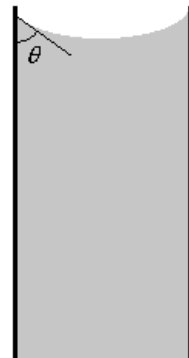
Forces on a molecule of liquid

Liquid in a vertical tube

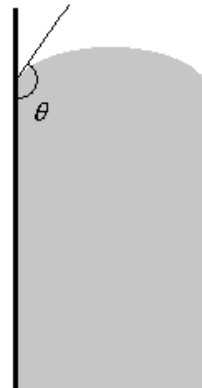
$$h = \frac{2\gamma_{la} \cos \theta}{\rho g r}$$



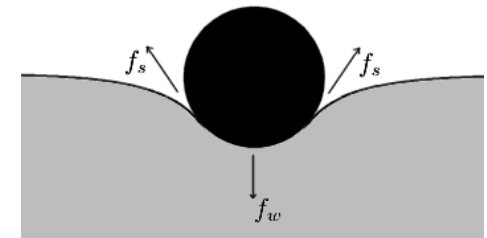
Hg barometer



Concave meniscus



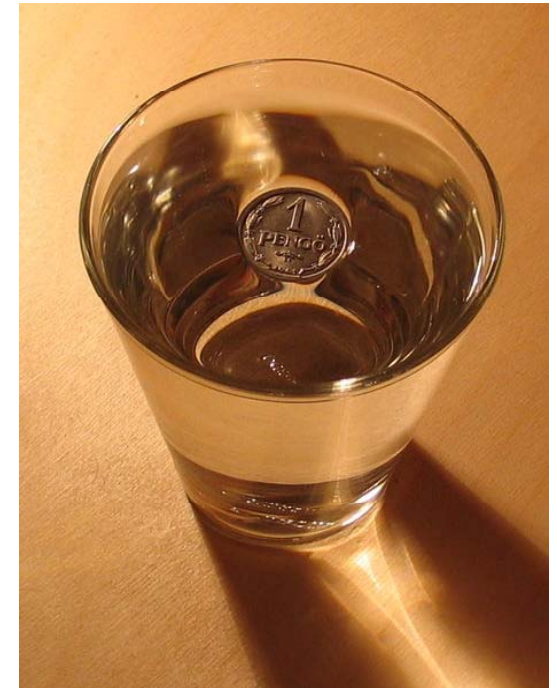
Convex meniscus



A needle floating on the surface of water

Surface Tension-Powered Self-Assembly

Surface tension in everyday life

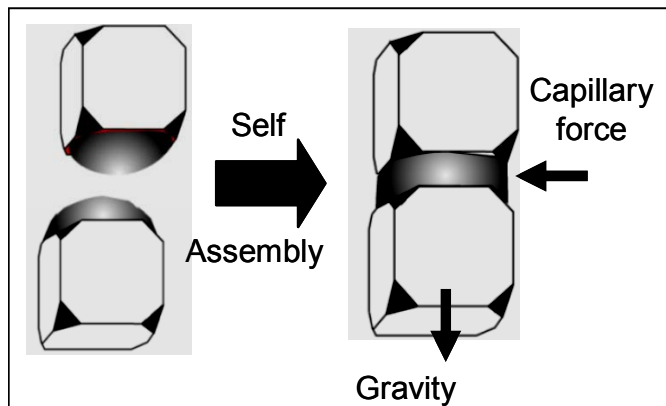


Surface Tension-Powered Self-assembly

Surface tension is dominating at micro and nano-scale

Why surface tension force?

- Surface tension forces are very strong on the nanoscale
- Roughness not a problem: Large scale integration possible



Scaling law

$$\frac{\text{Surface Tension Based Force}}{\text{Gravity (volume)}} \approx \frac{r}{r^3} = \frac{1}{r^2}$$

e.g. for a 50 nm cube of copper coated with solder faces,

Solder (Sn/Pb): 542 dynes/cm
(H₂O: 72 dynes/cm at 25°C)

$$F(\text{capillary}) = 2\pi r\gamma \approx 1 \times 10^{-7} \text{ N}$$

$$F(\text{gravity}) = mg = Vdg \approx 2 \times 10^{-17} \text{ N}$$

$$F(kT) \approx 1 \times 10^{-13} \text{ N} \quad (\text{Thermal force})$$

Surface Tension-Powered Self-Assembly

Two-dimensional geometry for surface tension powered rotation.

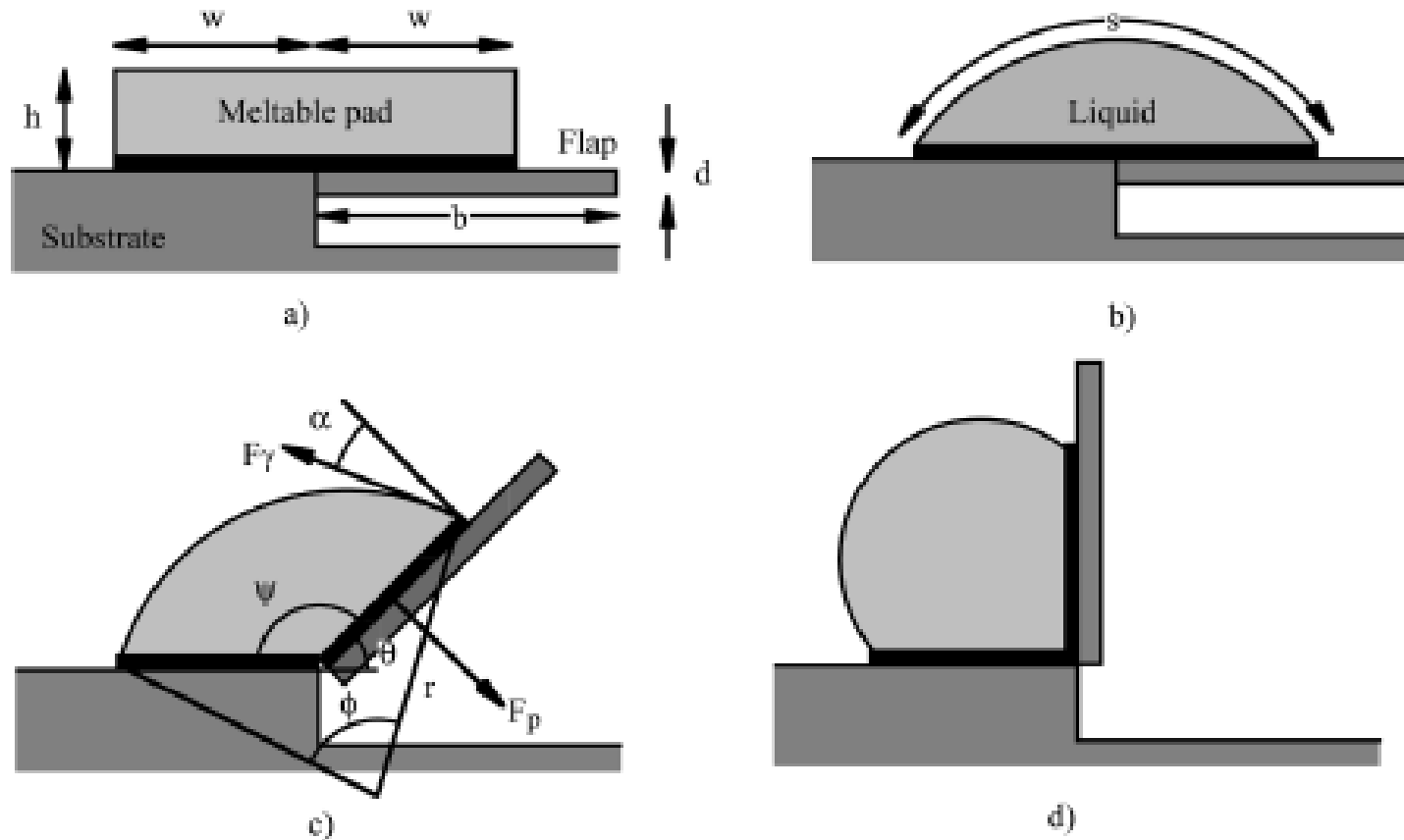
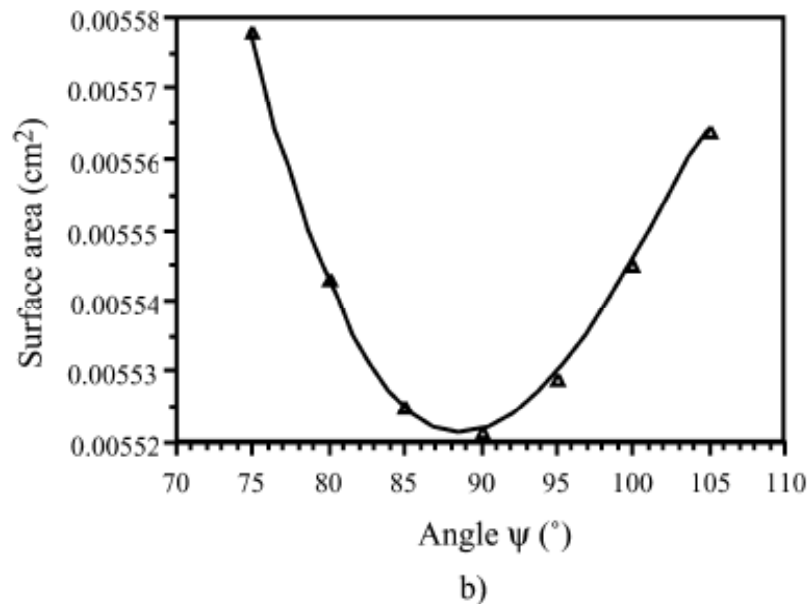
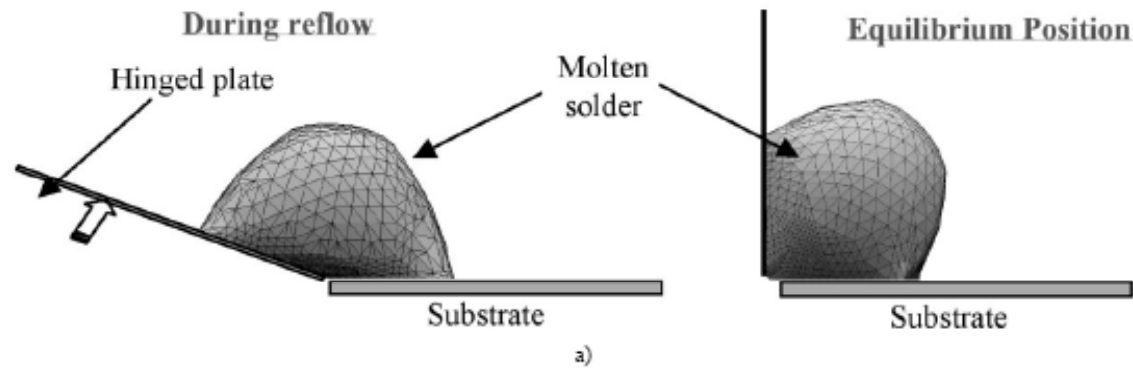


Fig. 3. Two-dimensional geometry for surface tension powered rotation.

Syms et al., 2003

Surface Tension-Powered Self-Assembly



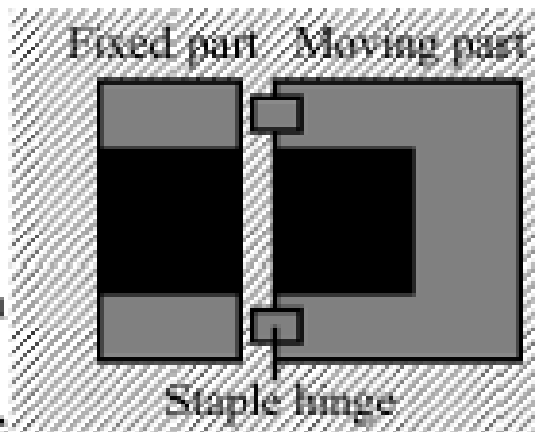
Syms et al., 2003

- a) FE simulation of surface tension powered rotation;
- b) variation of surface energy with angle, as predicted by the FE method

Surface Tension-Powered Self-Assembly

Example I

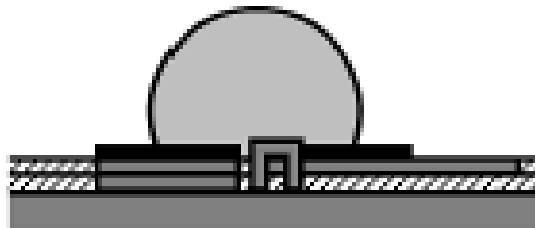
-  Solder
-  Gold
-  Poly-Si
-  Silica on silicon substrate



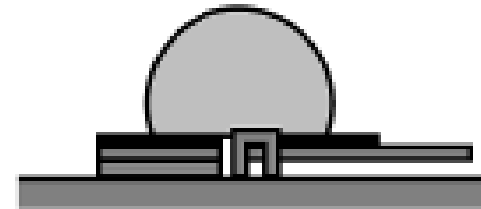
Fabricate hinged MUMPS structure



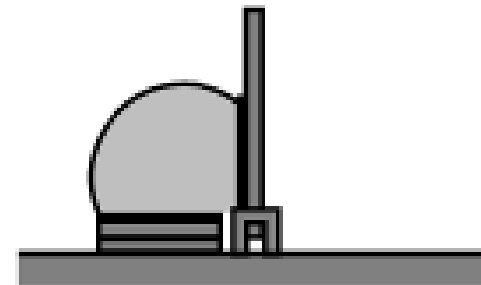
Place BGA Solder



Sacrificial etch

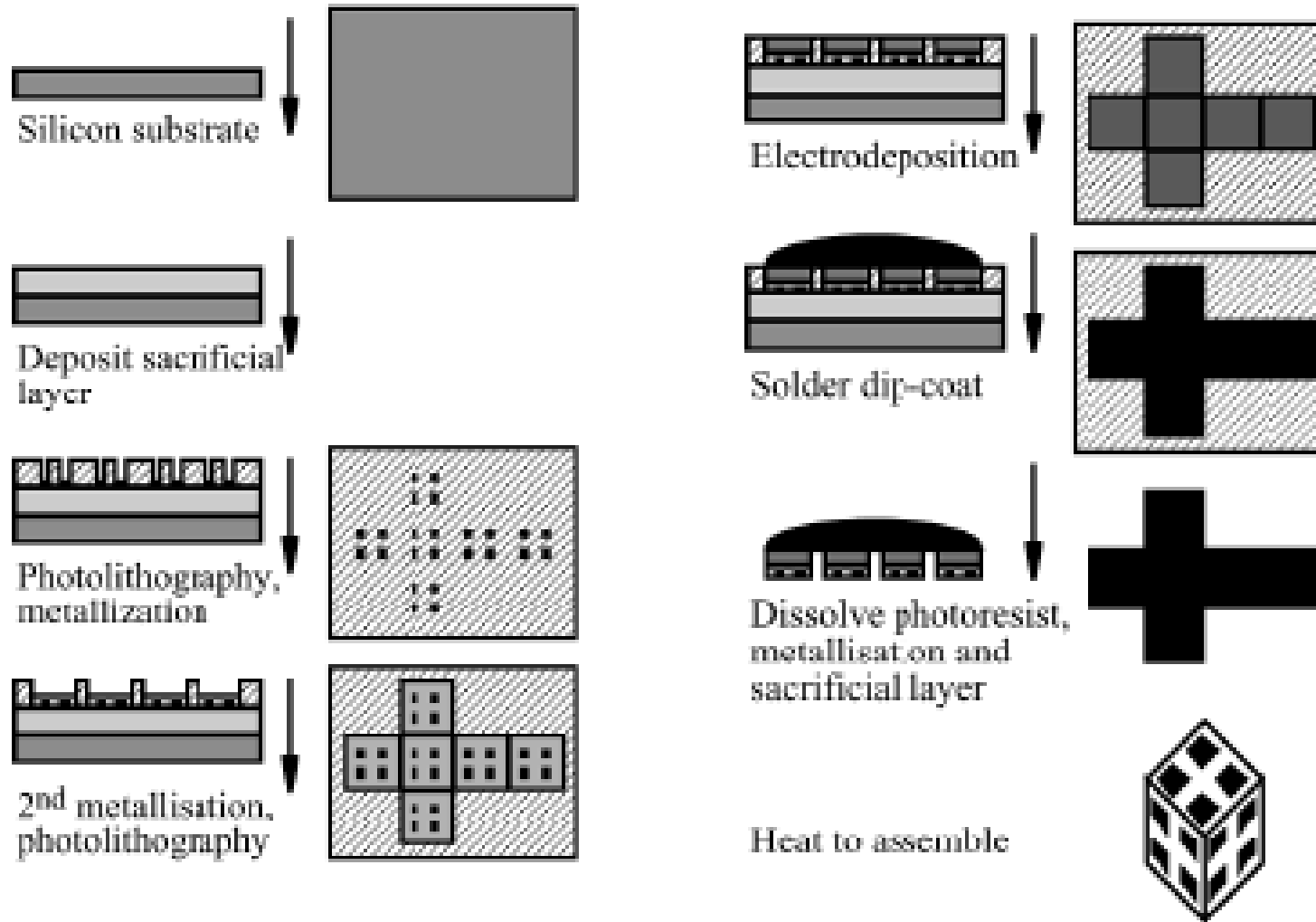


Melt solder



Surface Tension-Powered Self-Assembly

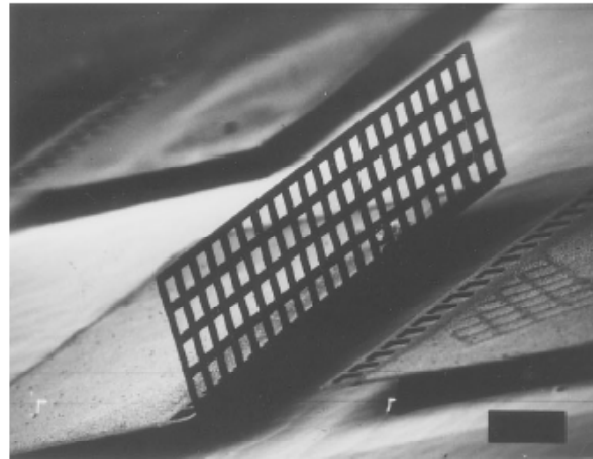
Example II



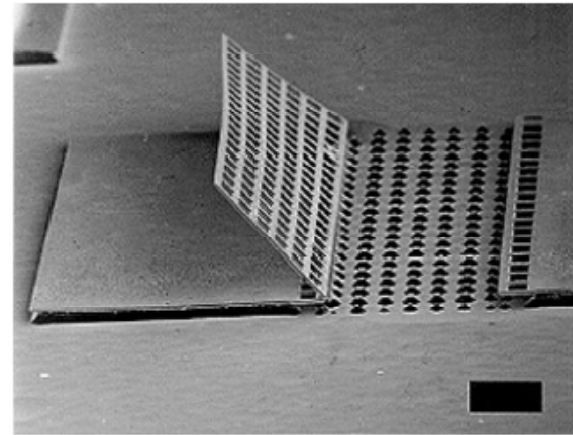
Syms et al., 2003

Surface Tension-Powered Self-Assembly

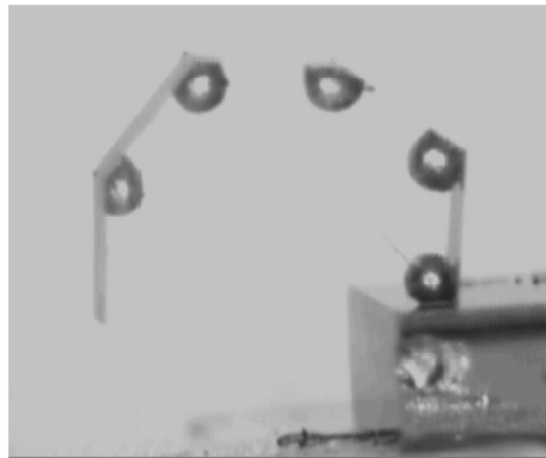
Some Examples



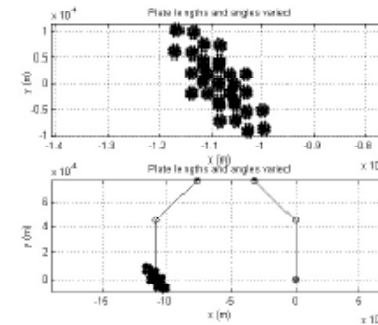
(a)



(b)



(c)



(d)

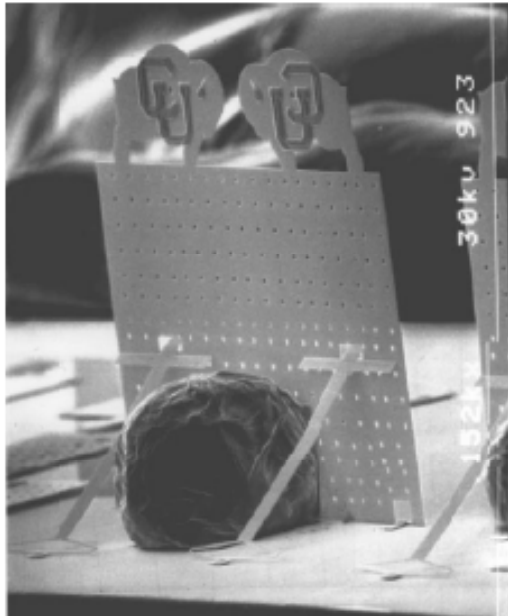
a) 90 rotation and b) over-rotation in hingeless structures; c) 5-bar multiple link assembly, and d) simulation of assembly accuracy.

Syms et al., 2003

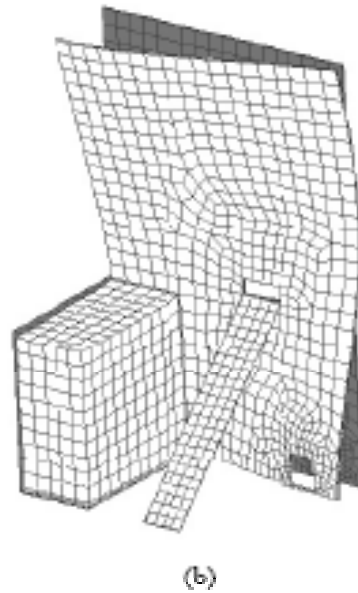
Self-assembly and Nanotechnology

Surface Tension-Powered Self-Assembly

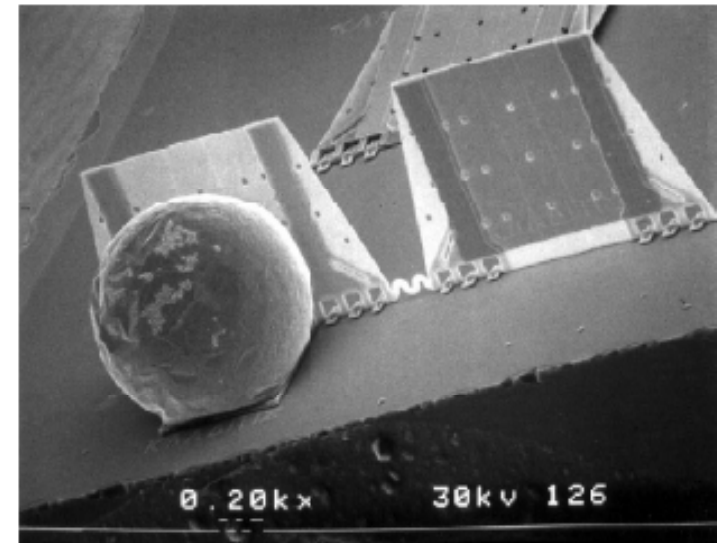
Some Examples



(a)



(b)



(c)

(a) Geometry and (b) predicted deformation of flap assembled using solder spheres; (c) mirrors assembled using solder self-assembly but attached via linkages.

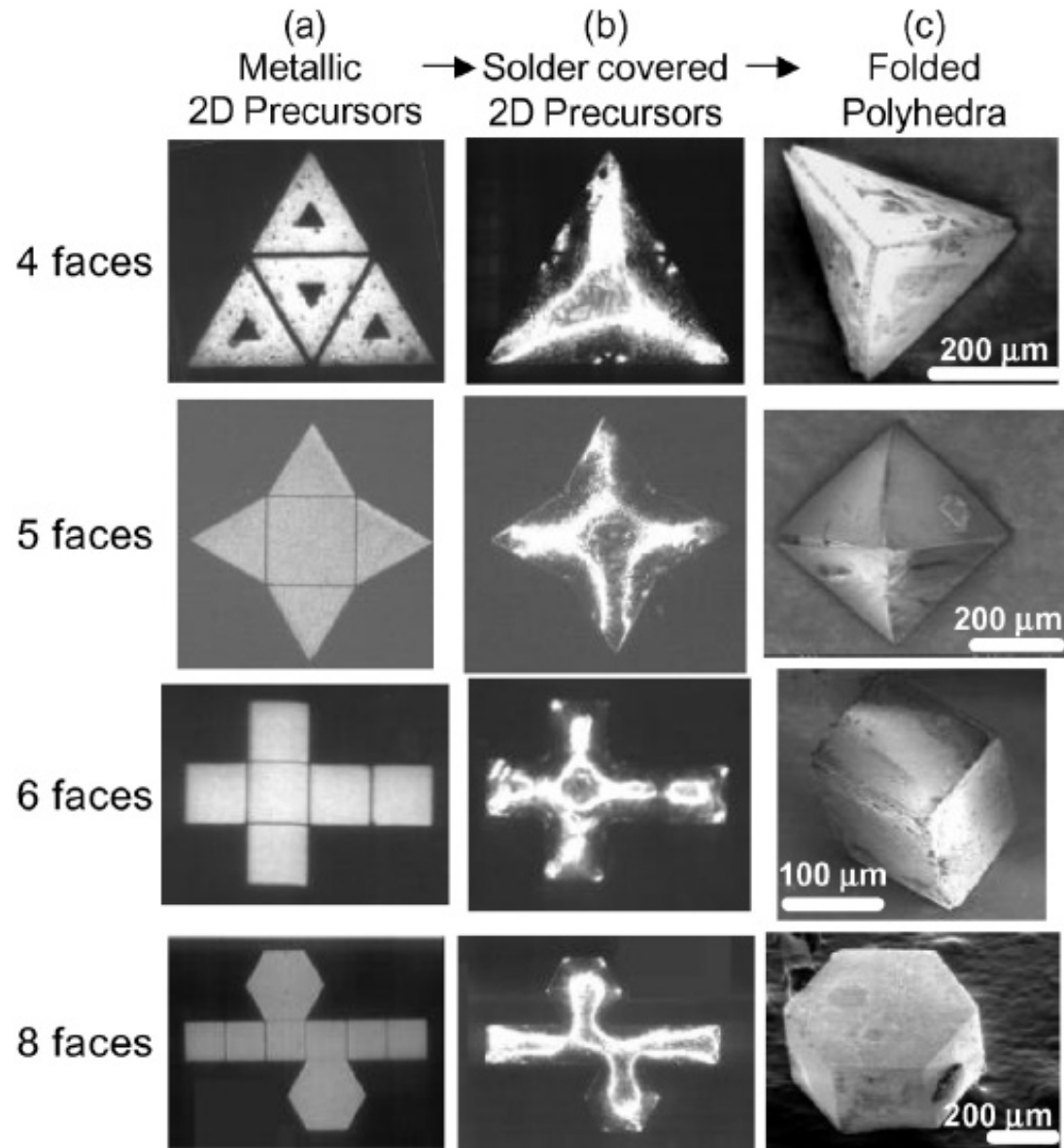
Syms et al., 2003

Surface Tension-Powered Self-Assembly

Some Examples

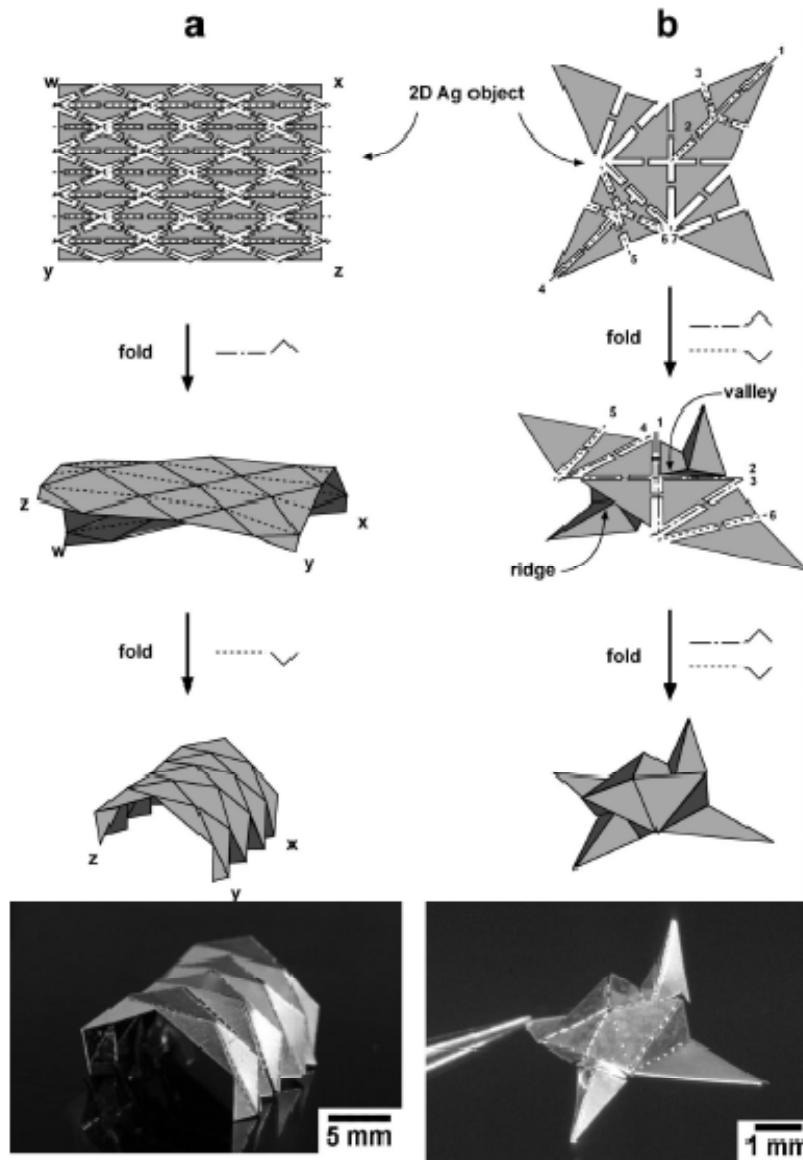
(a) Metallic 2-D structures prior to folding; (b) 2-D precursors after dip-coating with solder; (c) completed polyhedra, ranging in size from 100 to 300 μm per side

Syms et al., 2003



Surface Tension-Powered Self-Assembly

Some Examples

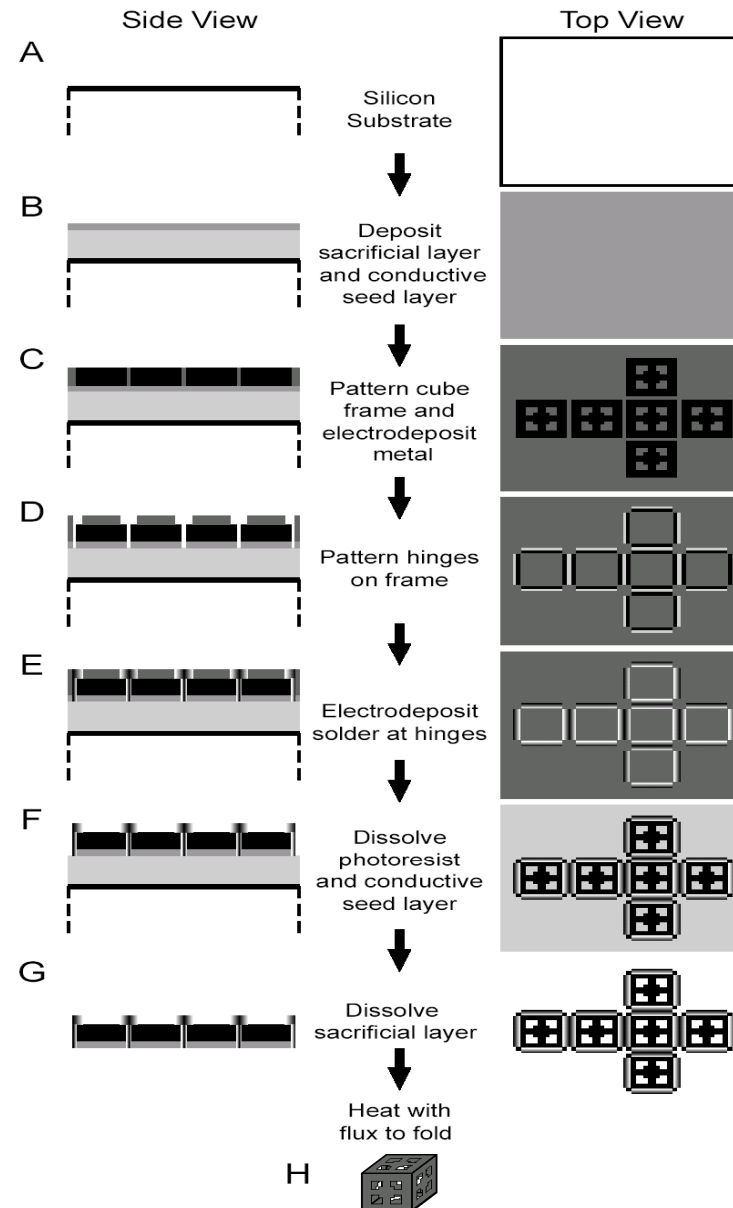


Syms et al., 2003

(a) Fold patterns and (b) completed structures for two micro-origami figures.

Case Study: Nano-liter Boxes for Encapsulation and Release

Fabrication and self-assembly

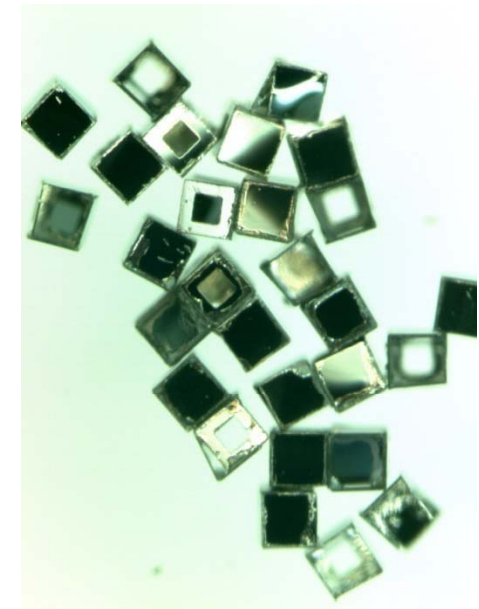
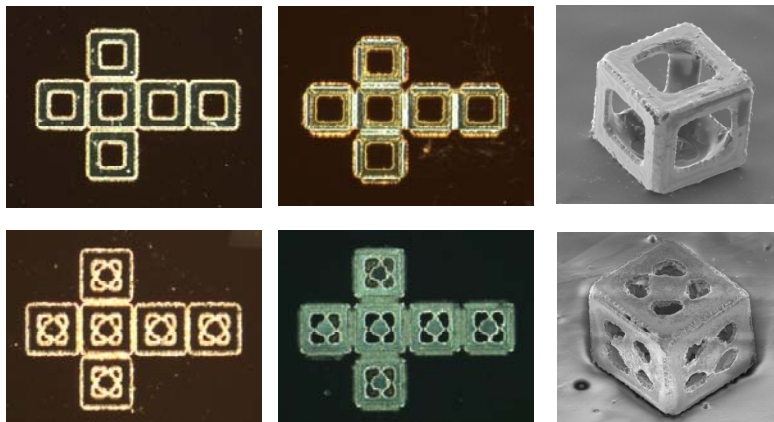


Gimi et al., Biomedical Microdevices, 2005

Self-assembly and Nanotechnology

Case Study: Nano-liter Self-assembled Boxes

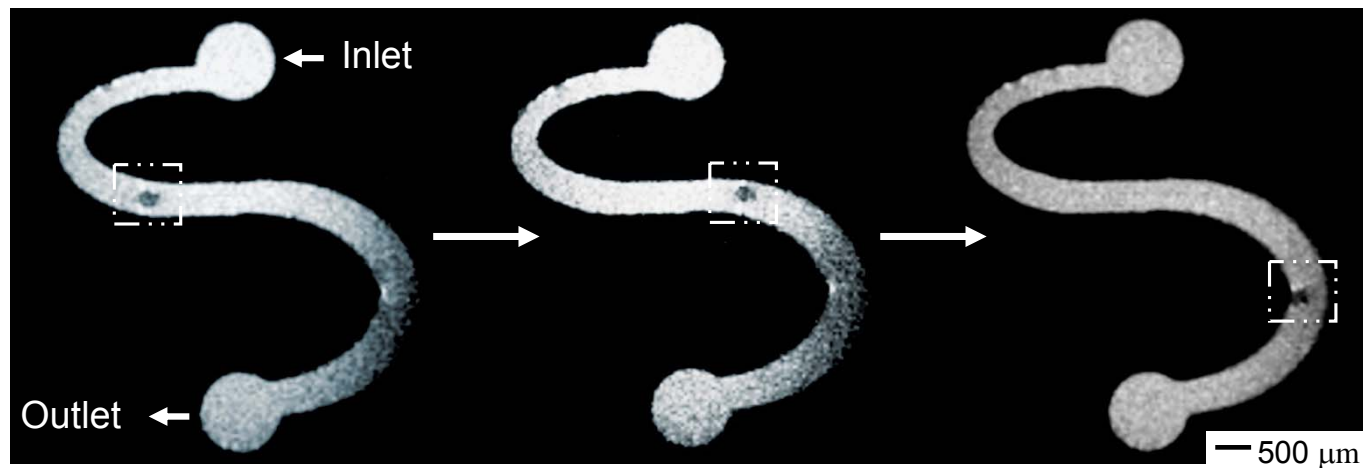
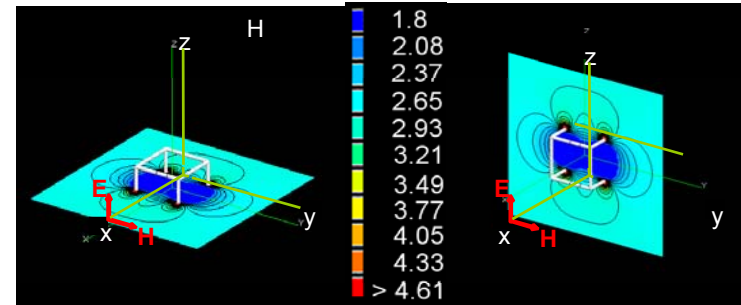
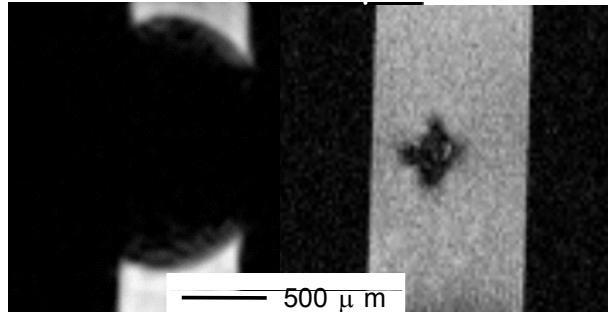
Fabrication and self-assembly



Gimi et al., Biomedical Microdevices, 2005

Case Study: Nano-liter Self-assembled Boxes

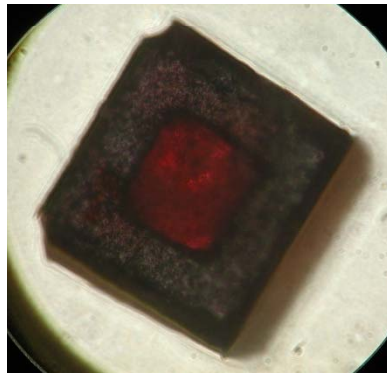
Magnetic Resonance Imaging (MRI) tracking



Gimi et al., Biomedical Microdevices, 2005

Case Study: Nano-liter Self-assembled Boxes

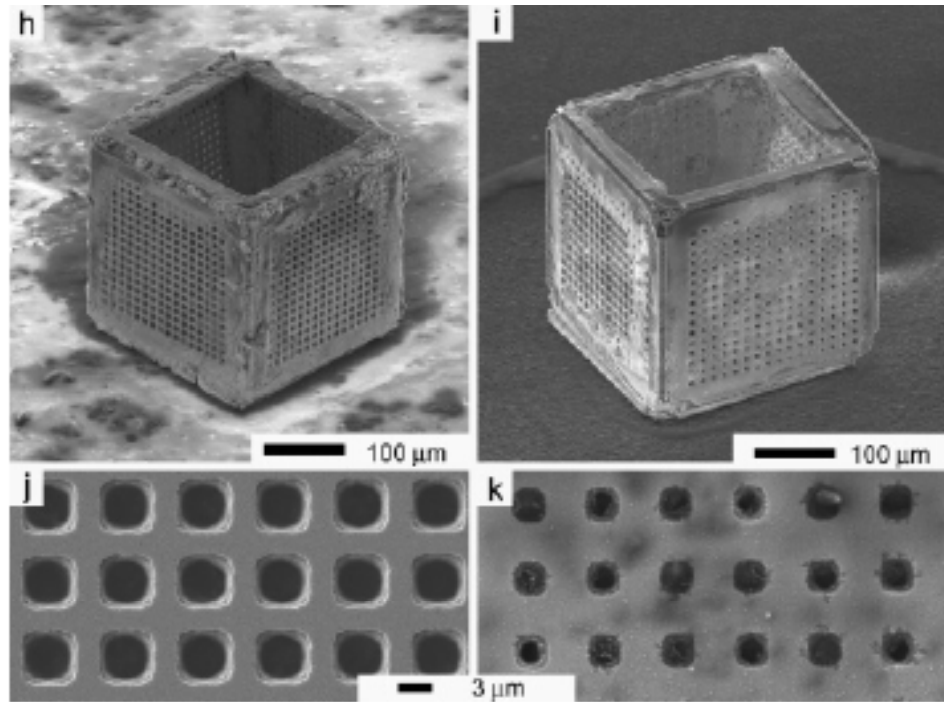
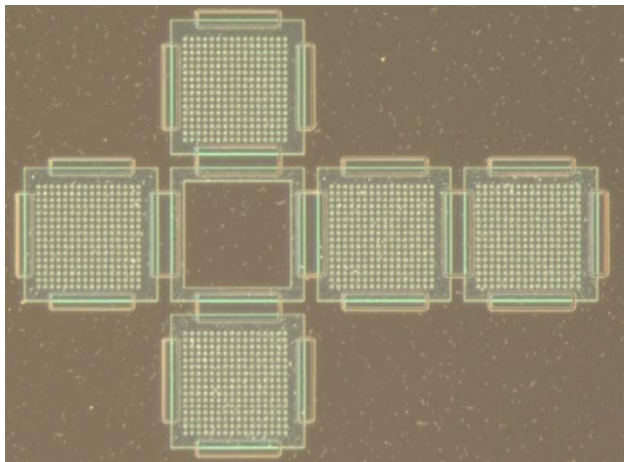
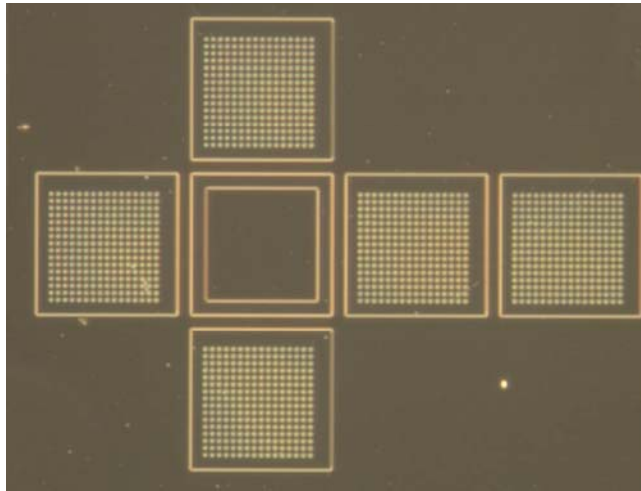
Particle and cell loading and controlled release



Gimi et al., Biomedical Microdevices, 2005

Case Study: Nano-liter Self-assembled Boxes

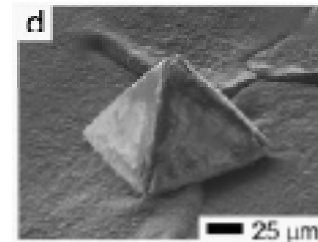
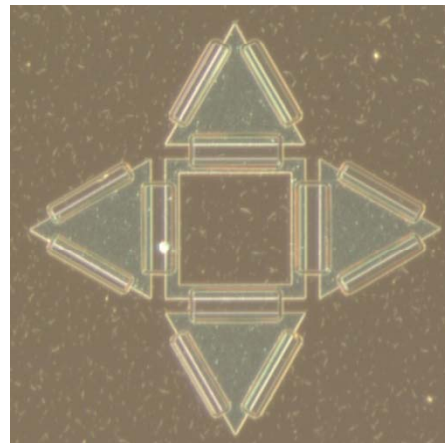
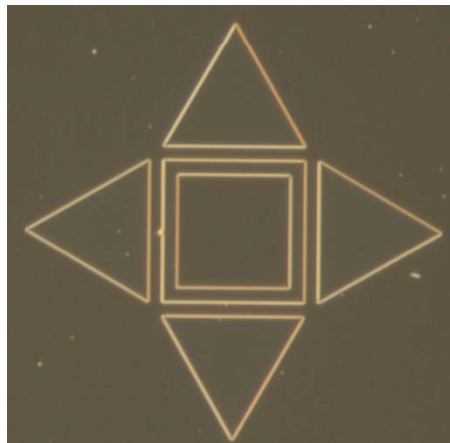
Controlling the porosity



Leong et al., JACS, 2006

Case Study: Nano-liter Self-assembled Boxes

Controlled the shapes

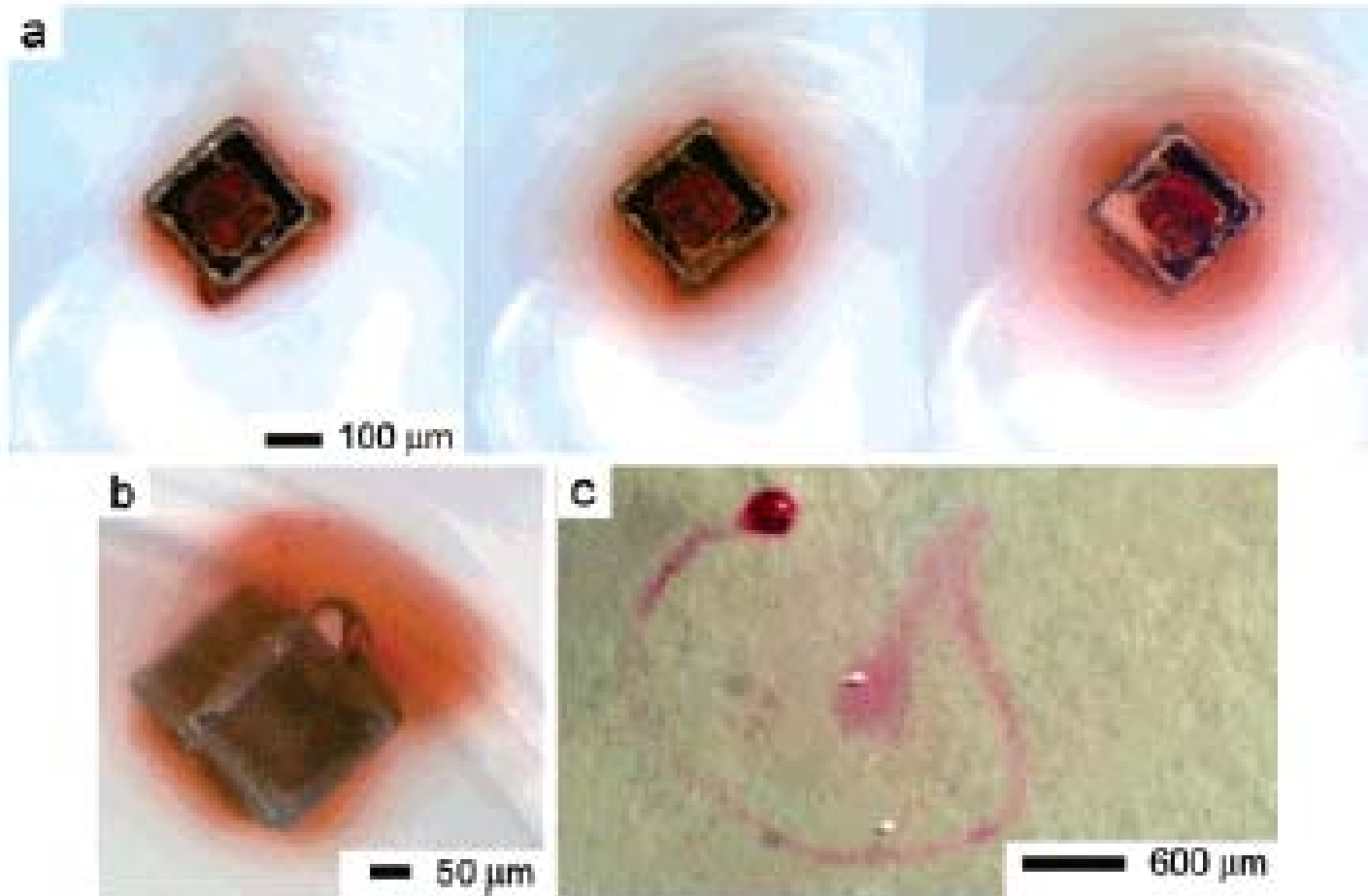


Leong et al., JACS, 2006

Self-assembly and Nanotechnology

Case Study: Nano-liter Self-assembled Boxes

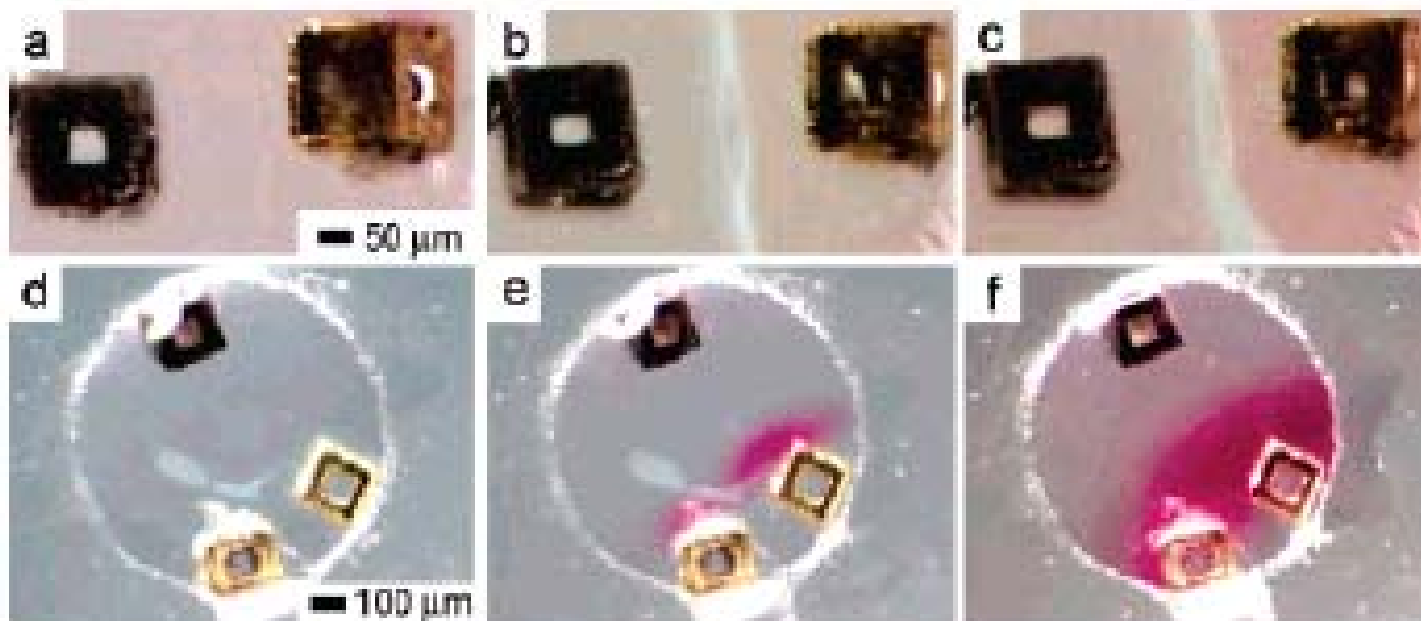
Controlled diffusion and release profiles



Leong et al., JACS, 2006

Case Study: Nano-liter Self-assembled Boxes

Controlled reactions



(a-c) reaction of copper sulfate and potassium hydroxide in an aqueous medium resulting in the formation of copper hydroxide along the central line between the containers; (d-f) the reaction of phenolphthalein (diffusing out of the two bottom containers) and potassium hydroxide (diffusing out of the top container) in an aqueous medium.

Leong et al., JACS, 2006

Lecture 2: Fundamentals and Theories of Self-Assembly

Final thought: Create or develop self-assembled structures at different scales



Atoms, molecules, ●●●



Human, earth, ●●●