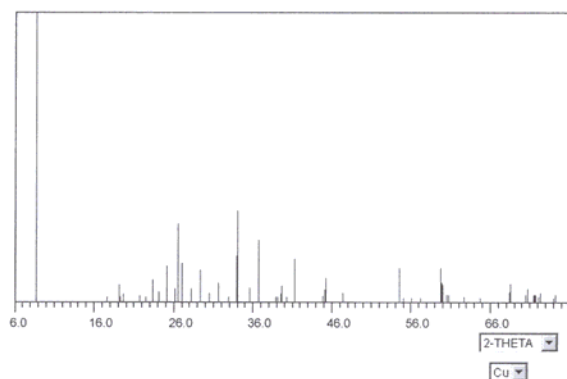


XRD REFERENCE SPECTRA

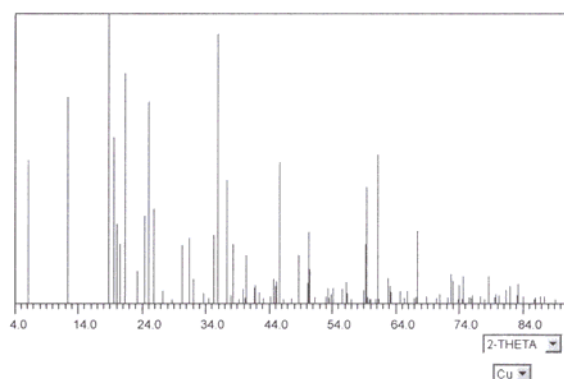
All spectra are for Cu-K α x-rays ($\lambda = 1.54\text{\AA}$). $\text{\AA} = \text{Angstrom, a unit of distance} = 1 \times 10^{-10} \text{ m}$. I/I° is the relative intensity. $I/I^\circ = 100$ identifies the diffracted peak with the greatest intensity. The y-axis on each of the images is the relative intensity (I/I°). The line that extends to the top of the graph represents the peak with the greatest intensity.

CLAYS IN POTTERY

BIOTITE, [3], 2M(1), $K(\text{Mg,Fe})_3\{\text{AlSi}_3\text{O}_{10}\}(\text{OH})_2$



CHLORITE, [1], $\text{Al}_{2.0}[\text{Si}_{3.3}\text{Al}_{0.7}]\text{O}_{10}(\text{OH})_2[\text{Mg}_{2.3}\text{Al}_{0.7}](\text{OH})_6$



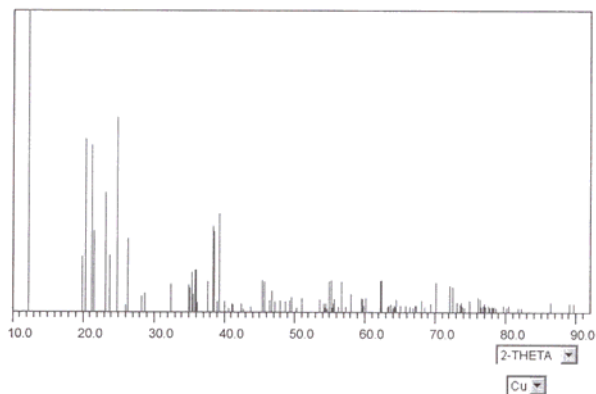
Biotite

| d | I/I° | θ | 2 θ |
|--------|-------------|----------|------------|
| 10.064 | 100 | 4.4 | 8.8 |
| 2.624 | 44 | 17.1 | 34.1 |
| 3.401 | 36 | 13.1 | 26.2 |

Chlorite

| d | I/I° | θ | 2 θ |
|-------|-------------|----------|------------|
| 4.728 | 100 | 9.4 | 18.8 |
| 2.503 | 92 | 17.9 | 35.8 |
| 4.172 | 79 | 10.6 | 21.3 |

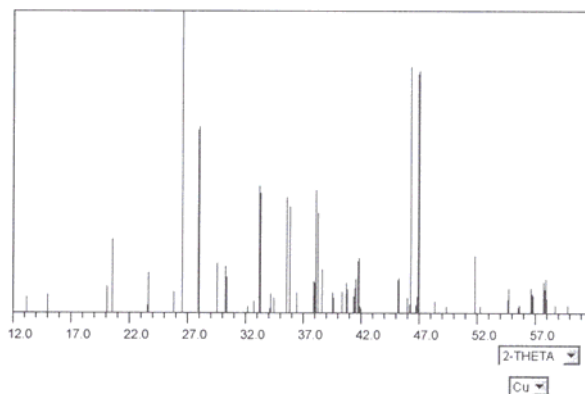
KAOLINITE, [2], $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$



Kaolinite

| d | I/I° | θ | 2 θ |
|-------|------|----------|------------|
| 8.886 | 100 | 5.0 | 10.0 |
| 5.425 | 12 | 8.2 | 16.3 |
| 4.643 | 9 | 9.5 | 19.1 |

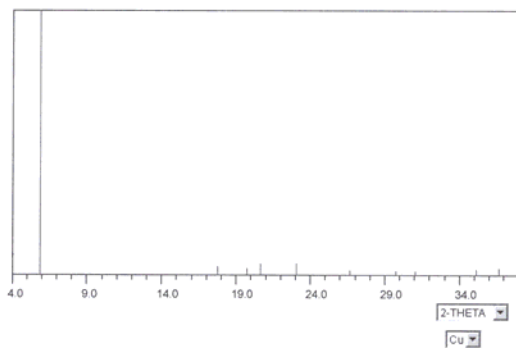
KYANITE, [1], structure type - kyanite, $\text{Al}_2[\text{SiO}_5]$



Kyanite

| d | I/I° | θ | 2 θ |
|-------|------|----------|------------|
| 3.345 | 100 | 13.3 | 26.6 |
| 1.957 | 81 | 23.2 | 46.3 |
| 1.929 | 80 | 23.5 | 47.0 |

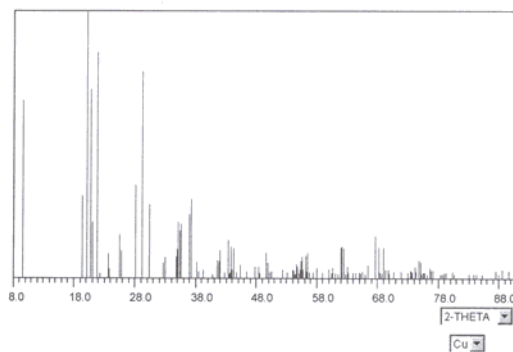
MONTMORILLONITE, [1], Ca, syn, DE, $\text{Ca}_{0.5}\text{Al}_2\text{Si}_4\text{O}_{10}(\text{OH})_2 \cdot \text{H}_2\text{O}$



Montmorillonite

| d | I/I° | θ | 2 θ |
|--------|------|----------|------------|
| 15.000 | 100 | 2.9 | 5.8 |
| 4.299 | 4 | 10.3 | 20.6 |
| 3.851 | 4 | 11.5 | 23.0 |

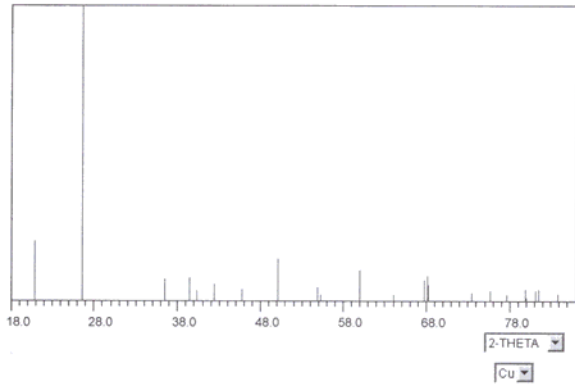
PYROPHYLLITE, [1], ITC, $\text{Al}[\text{Si}_2\text{O}_5](\text{OH})$



Pyrophyllite

| d | I/I° | θ | 2 θ |
|-------|------|----------|------------|
| 4.411 | 100 | 10.1 | 20.1 |
| 4.060 | 84 | 10.9 | 21.8 |
| 3.061 | 77 | 14.6 | 29.0 |

QUARTZ, [1], alpha, structure type - alpha-quartz, SiO₂

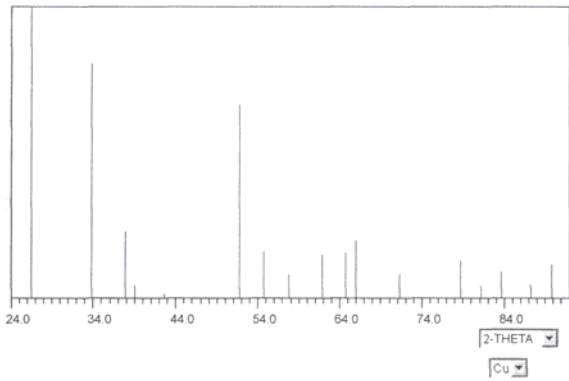


Quartz

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 3.344 | 100 | 13.3 | 26.6 |
| 4.256 | 20 | 10.4 | 20.8 |
| 1.818 | 14 | 25.1 | 50.1 |

POTTERY GLAZES

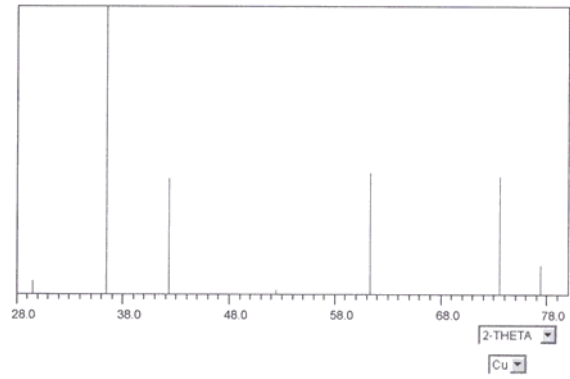
CASSITERITE, [1], structure type - rutile, SnO₂



Cassiterite

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 3.350 | 100 | 13.3 | 26.6 |
| 2.644 | 80 | 16.9 | 33.8 |
| 1.764 | 66 | 25.9 | 51.8 |

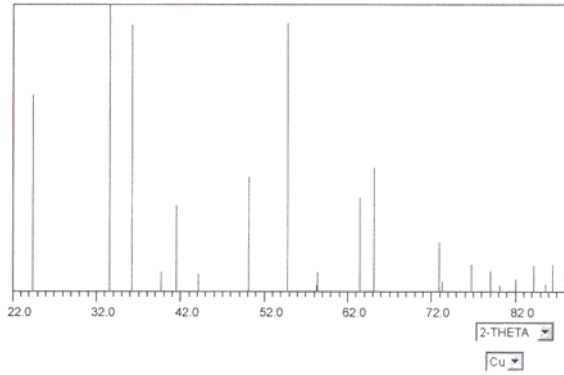
CUPRITE, [1], structure type - cuprite, Cu₂O



Cuprite

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 2.465 | 100 | 18.2 | 36.4 |
| 1.510 | 41 | 30.7 | 61.3 |
| 1.287 | 41 | 36.7 | 73.5 |

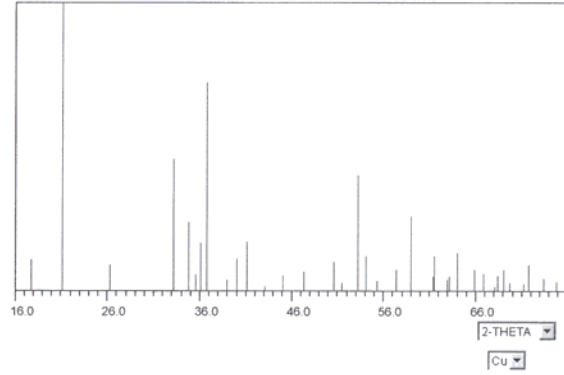
ESKOLAITE, [1], structure type - corundum, Cr₂O₃



Eskolaite

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 2.666 | 100 | 16.8 | 33.6 |
| 1.673 | 93 | 27.4 | 54.8 |
| 2.480 | 93 | 18.1 | 36.2 |

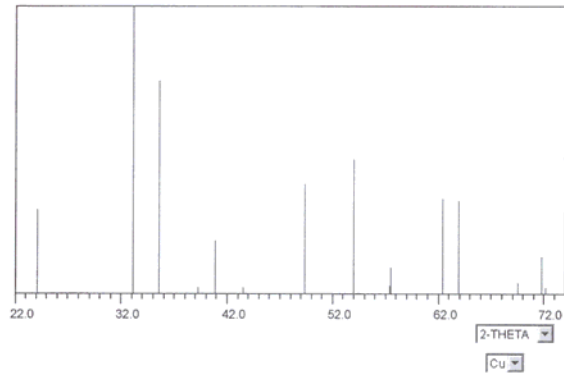
GOETHITE, [1], alpha, structure type - diaspore, FeO(OH)



Goethite

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 4.190 | 100 | 10.6 | 21.2 |
| 2.445 | 72 | 18.4 | 36.7 |
| 2.694 | 45 | 16.6 | 33.2 |

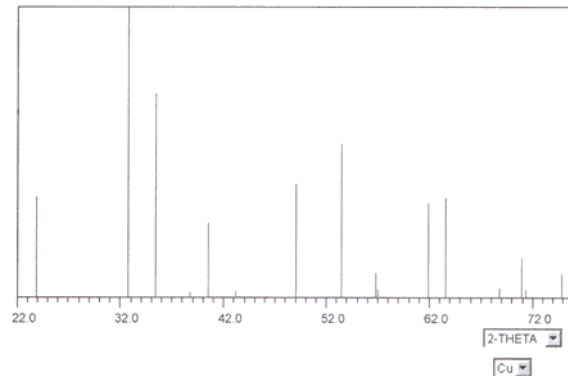
HEMATITE, [1], structure type - corundum, Fe₂O₃



Hematite

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 2.703 | 100 | 16.6 | 33.1 |
| 2.519 | 74 | 17.8 | 35.6 |
| 1.697 | 46 | 27.0 | 54.0 |

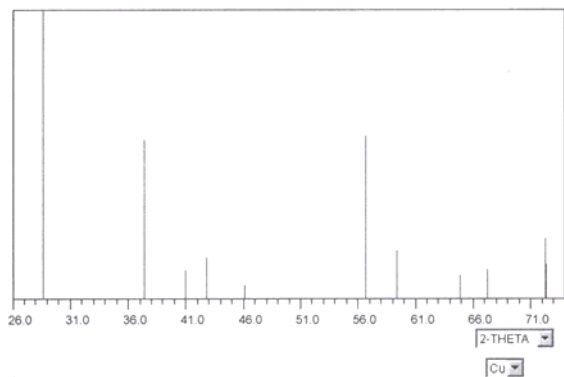
ILMENITE, [1], structure type - ilmenite, Fe(Ti,Mg)O₃



Ilmenite

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 2.728 | 100 | 16.4 | 32.8 |
| 2.534 | 70 | 17.7 | 35.4 |
| 1.712 | 52 | 26.7 | 53.4 |

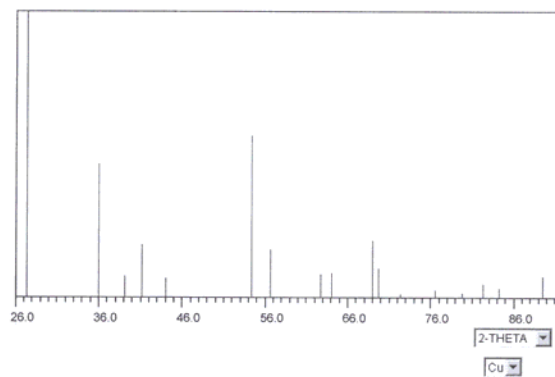
PYROLUSITE, [1], MnO₂



Pyrolusite

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 3.110 | 100 | 14.3 | 28.6 |
| 1.623 | 56 | 28.3 | 56.6 |
| 2.405 | 55 | 18.7 | 37.4 |

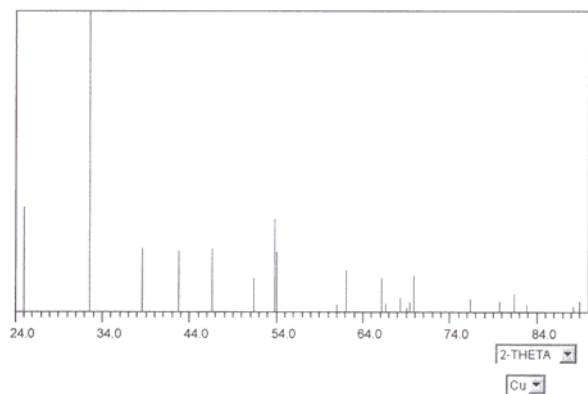
RUTILE, [1], structure type - rutile, at 25°C, TiO₂



Rutile

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 3.248 | 100 | 13.7 | 27.4 |
| 1.687 | 56 | 27.2 | 54.3 |
| 2.487 | 46 | 18.0 | 36.1 |

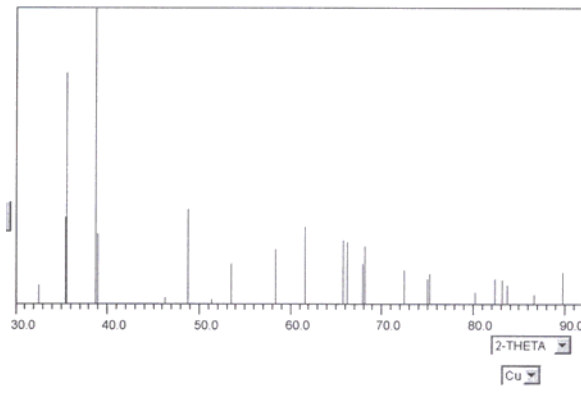
SPHAEROCOBALTITE, [1], CoCO₃



Sphaerocobaltite

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 2.742 | 100 | 16.3 | 32.6 |
| 3.550 | 34 | 12.5 | 25.0 |
| 1.702 | 30 | 26.9 | 53.8 |

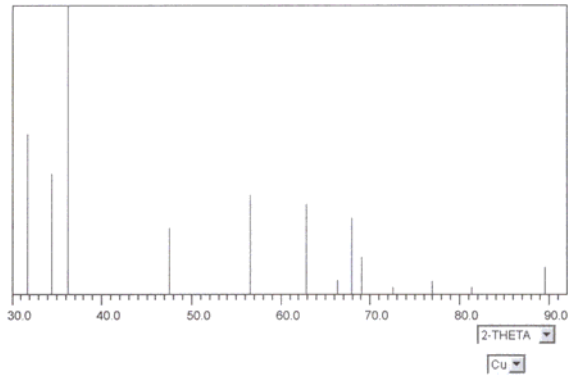
TENORITE, [1], structure type - tenorite, CuO



Tenorite

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 2.322 | 100 | 19.4 | 38.7 |
| 2.523 | 78 | 17.8 | 35.6 |
| 1.866 | 31 | 24.4 | 48.8 |

ZINCITE, [5], ZnO

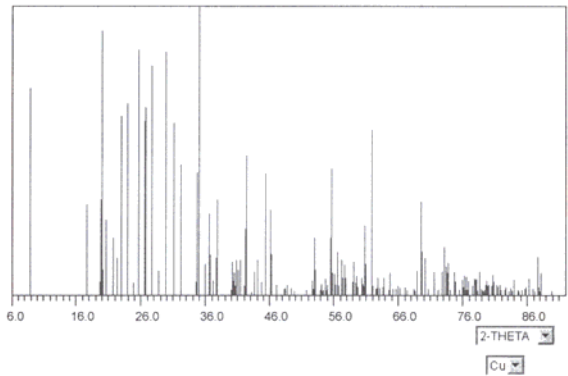


Zincite

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 2.475 | 100 | 18.1 | 36.2 |
| 2.814 | 55 | 15.9 | 31.8 |
| 2.602 | 41 | 17.3 | 34.4 |

COSMETICS

MUSCOVITE, [1], 2M, $KAl_2[Si_3Al]O_{10}(OH)_2$



Muscovite is used in some cosmetics to provide a pearly-like appearance. Other minerals used in modern cosmetics are rutile and hematite for yellowish and reddish colors respectively. XRD data for these two minerals are found in the preceding section for “Pottery Glazes”.

Muscovite

| d | I/I° | θ | 2θ |
|-------|------|------|------|
| 2.557 | 100 | 17.5 | 35.0 |
| 4.448 | 91 | 10.0 | 20.0 |
| 3.476 | 85 | 12.8 | 25.6 |